REVIEW: Chapter 8 - Covalent Bonding

1. Typically, how many bonds & lone pairs will each have when covalently bonding?

a. Boron - 3 bonds: 0 lone pairs

d. lodine - 1 bond; 3 lone pairs

b. Silicone - 4 bonds; O lone pairs

e. Nitrogen - 3 bonds; 1 lone pair

c. Neon - 0 bonds; 4 lone pairs

f. Oxygen - 2 bonds; 2 lone pairs

2. Which elements do not follow the octet rule? (Hint - there are four we went over)

a. __H__ & __B__ never follow the octet rule.

b. __P__ & __5_ sometimes don't follow the octet rule.

3. What type of bond will each combination of atoms form?

a. F & N = 2 nonmetals, therefore covalent

d. B & CI - 2 nonmetals, therefore covalent

b. Al & O - 1 metal/1 nonm., therefore ionic

e. Ca & I - 1 metal/1 nonm., therefore ionic

c. Na & 5 - 1 metal/1 nonm., therefore ionic

f. Fe & Cu - 2 metals, therefore metallic

- 4. What is the difference between:
 - a. Molecules & Compounds?

Compounds can be made using all types of bonds (ionic, covalent or metallic), but molecules can only be made using covalent bonds

b. Monatomic elements & Diatomic molecules ?

Monatomic elements exist as single atoms; diatomic molecules are made up of two of the same atom

c. Structural formulas, Empirical formulas, & Molecular formulas?

All three are for covalent bonds only! Molecular show the actual number of atoms in the molecule; Empirical give the lowest whole number ratio; Sturctural formulas are not really formulas, they are drawing of the molecule.

d. Polar bonds & Non-polar bonds?

Both describe covalent bonds! In a polar bond the electrons are NOT shared equally; in a non-polar bond electrons are shared equally

5. Classify each of the following as an atom or a molecule:

a. Be, atom

b. CO, molec.

c. Na molec.

d. H₂O, molec.

e. Ne. atom

Section 8 Covalent Bonding Workbook Answers

Daniel C. Tofan

Section 8 Covalent Bonding Workbook Answers:

Chemistry Workbook For Dummies Peter J. Mikulecky, Katherine Brutlag, Michelle Rose Gilman, Brian Peterson, 2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence Investigating Chemistry Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package E3 Chemistry Guided Study Book - 2018 Home Edition (Answer Key Included) Effiong Eyo, 2017-12-08 Chemistry students and Homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Guided Study Book 2018 With E3 Chemistry Guided Study Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with guided step by step solutions to study and follow Practice multiple choice and short answer questions along side each concept to immediately test student understanding of the concept 12 topics of Regents question sets and 2 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 1979088374 The Home Edition contains answer key to all questions in the book Teachers who want to recommend our Guided Study Book to their students should recommend the Home Edition Students and and parents whose school is not using the Guided Study Book as instructional material as well as homeschoolers should also buy the Home edition The School Edition does not have the answer key in the book A separate answer key booklet is provided to teachers with a class order of the book Whether you

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Science for Tenth Class Part 2 Chemistry Lakhmir Singh & Manjit Kaur, A series of books for Classes IX and X according to the CBSE syllabus and CCE Pattern

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