

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the processes of a reaction.
- 2. Explain how to write a balanced equation.
- 3. List the steps for writing a complete ionic equation.

Vocabulary

- 1. chemical equation
- 2. balanced equation
- 3. reactant
- 4. coefficient
- 5. product

Part A Completion

Write the appropriate word(s) in space or under the blank line, and underline the word(s) in the correct definition. Write complete Latin and Greek names in italics.

A chemical equation is usually represented by a symbol _____. The chemical formula for a chemical compound is _____. The law of conservation of mass states that the _____ of matter is conserved. The law of conservation of _____ states that the total number of _____ is conserved. The law of conservation of energy states that the _____ of energy is conserved. The law of conservation of mass states that the _____ of mass is conserved. The law of conservation of energy states that the _____ of energy is conserved. The law of conservation of mass states that the _____ of mass is conserved.

Chemical reactions are usually represented by a symbol _____. The chemical formula for a chemical compound is _____. The law of conservation of mass states that the _____ of matter is conserved. The law of conservation of _____ states that the total number of _____ is conserved. The law of conservation of energy states that the _____ of energy is conserved. The law of conservation of mass states that the _____ of mass is conserved.

1. equation
2. formula
3. conservation
4. energy
5. mass
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