



CHARACTERISTICS OF GAS

- **No Fixed Shape:** Fills the container it occupies evenly.
- **Expands Easily:** Occupies the entire available volume.
- **Low Density:** Particles are widely spaced and move freely.
- **Compressible:** Easily compressed into smaller volumes.
- **Rapid Diffusion:** Particles mix and spread quickly.
- **No Fixed Volume:** Takes the shape of the container.
- **Exerts Pressure:** Due to particle collisions with container walls.
- **High Kinetic Energy:** Particles have significant motion and energy.
- **No Attractive Forces:** Weak intermolecular forces between particles.
- **Turns into Liquid or Solid:** Under reduced temperature or pressure.

Section Characteristics Of Gases

D Keegan



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