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Student Exploration: Balancing Chemical Equations

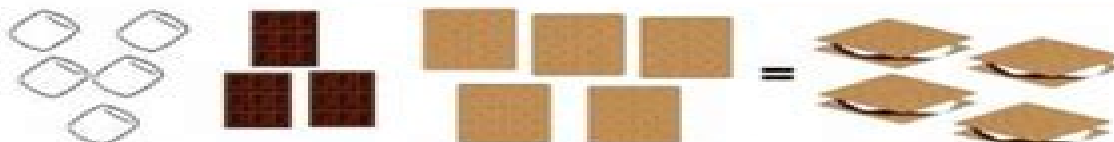
Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

The scouts are making s'mores out of toasted marshmallows, chocolate, and graham crackers.

1. What is wrong with the image below?



2. Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, how many s'mores could you make with the ingredients shown?

Gizmo Warm-up

In a chemical reaction, **reactants** interact to form **products**. This process is summarized by a chemical equation. In the *Balancing Chemical Equations* Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$.

1. How many atoms are in a hydrogen **molecule** (H_2)?
2. How many atoms are in an oxygen molecule (O_2)?
3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)?
4. In general, what does a **subscript** (such as the "2" in H_2) tell you about the molecule?



5. A chemical equation is **balanced** if the number of each type of atom on the left side is equal to the number of each type on the right side. Is this reaction balanced?

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Brendan G. Carr



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