

**Activity B:****Methane reaction**Get the Gizmo ready:

- Select  $C + H_2$  becomes  $CH_4$ .



**Goal:** Predict the results of a chemical reaction.

1. Count:  $CH_4$  is the chemical formula for methane. In order to produce a single molecule of methane, how many carbon (C) atoms are needed?  Hydrogen atoms?

2. Predict: Set the number of C atoms to 14 and the number of  $H_2$  molecules to 10.

A. How many total hydrogen atoms are there?

B. How many methane molecules could form from these reactants?

C. After the atoms react, which reactant will be left over?

D. Which reactant will be the limiting reactant?

E. Click **Play** and wait until **Reaction complete** is shown. What happened?

Each hydrogen molecule formed a connection with carbon (4 hydrogen molecules per carbon). There are extra carbon atoms that did not link because there are additional carbon atoms.

3. Test: Click **Reset**. Set the number of C atoms to seven.

A. How many hydrogen molecules ( $H_2$ ) will be needed for there to be no excess reactants?  Use the slider to set the number of  $H_2$  molecules to this value.

B. How many methane molecules will be formed?

C. Click **Play**. What happened?

Since all of the hydrogen molecules linked to the carbon, there were no extra reactants.

4. Solve: For each **chemical equation** below, write the number of product molecules that will form from the reaction. Then, circle the limiting reactant. (Note: The **coefficients** in front of the reactants indicate the number of reactant molecules or atoms present.)



# Student Exploration Limiting Reactants Gizmo Answer Key

**Thirumalaisamy P.  
Velavancorresponding**



## **Student Exploration Limiting Reactants Gizmo Answer Key:**

## Adopting the Beat of Appearance: An Mental Symphony within **Student Exploration Limiting Reactants Gizmo Answer Key**

In a global taken by screens and the ceaseless chatter of immediate communication, the melodic elegance and psychological symphony created by the published term often disappear into the background, eclipsed by the constant sound and distractions that permeate our lives. Nevertheless, nestled within the pages of **Student Exploration Limiting Reactants Gizmo Answer Key** an enchanting fictional treasure full of natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by an elegant musician of language, this interesting masterpiece conducts visitors on an emotional trip, well unraveling the concealed tunes and profound influence resonating within each cautiously crafted phrase. Within the depths of the moving analysis, we shall examine the book is central harmonies, analyze its enthralling publishing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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