Name: Hannah Sonnentag Date: 12/31/11

Student Exploration: pH Analysis: Quad Color Indicator

Vocabulary: acid, acidic, alkaline, base, indicator, neutral, pH

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- Acids are substances that produce hydrogen ions (H*) when dissolved in water. Lemon
 juice is an example of an acid.
 - A. What does lemon juice taste like? SOUR
 - B. What does it feel like if lemon juice gets in your eye? burns
- Bases are substances that produce hydroxide ions (OH) when dissolved in water. Hand soap is an example of a base.
 - A. What does soap feel like? slippery, smooth
 - B. What does soap taste like? Sour, bad
 - C. What does it feel like if soap gets in your eye? burns

Gizmo Warm-up

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess H" ions there are in a solution. The pH scale runs from 0 to 14, with 0 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered **neutral**.

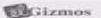
The pH Analysis: Quad Color Indicator Gizmo™ allows you to find the pH of a variety of liquids. In the Gizmo, check that the Substance in the tube is Ammonia, and click Test.



- Indicators change color in acids or bases. What are the colors of the pH paper?
 orange, coral pink, blue, yellow
- 2. Compare the paper to the pH color chart. What is the pH of ammonia?

 Is ammonia acidic or alkaline?

 alkaline



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- 1. Understanding the eBook Student Exploration Ph Analysis Answers
 - The Rise of Digital Reading Student Exploration Ph Analysis Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Student Exploration Ph Analysis Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Ph Analysis Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Student Exploration Ph Analysis Answers
 - Personalized Recommendations
 - Student Exploration Ph Analysis Answers User Reviews and Ratings
 - Student Exploration Ph Analysis Answers and Bestseller Lists
- 5. Accessing Student Exploration Ph Analysis Answers Free and Paid eBooks

- Student Exploration Ph Analysis Answers Public Domain eBooks
- Student Exploration Ph Analysis Answers eBook Subscription Services
- Student Exploration Ph Analysis Answers Budget-Friendly Options
- 6. Navigating Student Exploration Ph Analysis Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Student Exploration Ph Analysis Answers Compatibility with Devices
 - Student Exploration Ph Analysis Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Ph Analysis Answers
 - Highlighting and Note-Taking Student Exploration Ph Analysis Answers
 - Interactive Elements Student Exploration Ph Analysis Answers
- 8. Staying Engaged with Student Exploration Ph Analysis Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Ph Analysis Answers
- 9. Balancing eBooks and Physical Books Student Exploration Ph Analysis Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Student Exploration Ph Analysis Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Ph Analysis Answers
 - Setting Reading Goals Student Exploration Ph Analysis Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Ph Analysis Answers
 - Fact-Checking eBook Content of Student Exploration Ph Analysis Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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