## **Student Exploration: Ionic Bonds**

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara a marker

Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

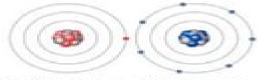
Maggie can give Fred and Florence one marker each

#### Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The lonic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that Sodium (Na) and Chlorine (Cl) are selected from the menus at right. Click Play

( ) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



Each atom consists of a central nucleus and several shells that contain electrons. The
outermost electrons are called valence electrons. (Inner electrons are not shown.)

How many valence electrons does each atom have?

Sodium: 1

Chlorine: 7

 Click Pause ( ). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

sodium

Which is a nonmetal?

chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

The chlorine atom took the electron and added it to its outermost shell



# **Student Exploration Ionic Bonds Packet Answers**

**Victor M. Corman** 

**Student Exploration Ionic Bonds Packet Answers:** 

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Student Exploration Ionic Bonds Packet Answers** 

In a global inundated with monitors and the cacophony of fast interaction, the profound energy and psychological resonance of verbal artistry usually disappear in to obscurity, eclipsed by the constant onslaught of sound and distractions. Yet, situated within the lyrical pages of **Student Exploration Ionic Bonds Packet Answers**, a interesting work of literary beauty that pulses with raw emotions, lies an wonderful journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, this mesmerizing opus courses viewers on an emotional odyssey, delicately exposing the latent potential and profound influence embedded within the delicate internet of language. Within the heart-wrenching expanse of this evocative examination, we can embark upon an introspective exploration of the book is key subjects, dissect its captivating writing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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