



Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Temperature and Particle Motion

**Essential Question:** What factors affect the temperature of a substance? How do these factors affect the motion of the particles in the substance?

**Prior Knowledge:** Speed and direction are both important in describing motion. ( )

1. Why is it not enough to say that an object is moving? Direction is also important. Velocity is the speed and direction of an object.
2. Why is it not enough to say that an object is moving? Speed is also important. Velocity is the speed and direction of an object.
3. An increase in the average speed of the particles in a substance is called temperature. If the average speed of the particles in a substance is 10 m/s, what is the average speed of the particles in a substance that is 20 m/s? 20 m/s. The average speed of the particles in a substance is directly proportional to the temperature.

**Warm-up:**

What happens when the temperature of a substance changes? The particles in the substance move faster or slower. The temperature of a substance is a measure of the average kinetic energy of the particles. The temperature of a substance is a measure of the average kinetic energy of the particles. The temperature of a substance is a measure of the average kinetic energy of the particles.

What happens when the temperature of a substance changes? The particles in the substance move faster or slower. The temperature of a substance is a measure of the average kinetic energy of the particles.

1. What happens when the temperature of a substance changes? The particles in the substance move faster or slower.

2. How does the temperature of a substance change? It changes when the average kinetic energy of the particles changes.



# Student Exploration Temperature And Particle Motion Answer Key

**Robert C. Smoot, Smoot, Richard G.  
Smith, Jack Price**



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