

13.2 Reactions Involving Acids and Bases

- Acids react with many metals.
- These reactions produce a solution of a metal salt and hydrogen gas.
- Salts are compounds usually made up of a metal cation and a non-metal anion.
- The salt produced will depend on the acid used in the reaction.
- Acids also react with many compounds such as metal hydroxides and metal carbonates.

Reaction Type 1: Acid + reactive metal \longrightarrow salt + hydrogen

- Reactive metals include Ca, Mg, K and Zn but not Cu, Ag or Au
- When dilute acids are added to main group metals and some transition metals, bubbles of hydrogen are released and a salt is formed.

The reaction between dilute hydrochloric acid and zinc metal.

This reaction can also be represented by an ionic equation.

Reaction Type 2: Acid + metal hydroxide \longrightarrow salt + water

- Metal hydroxides include NaOH, Ca(OH)₂ and Mg(OH)₂.
- The products of an acid-base reaction are a salt and water.

The reaction between sulphuric acid and sodium hydroxide.

The sulphuric acid is ionised in solution and both sodium hydroxide and sodium sulphate are ionic and therefore dissociated in solution. Water being a covalent molecular substance does not ionise to any significant extent. So the full equation is:

The ionic equation therefore is:

Study Guide Acid And Base Reactions

David R. Klein



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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 **Chemical Principles Student's Study Guide & Solutions Manual** John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text *Organic Chemistry Reactions: A Study Guide* Cybellium, 2024-10-26 Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you re a beginner or an advanced learner Cybellium books cover a wide range of topics from foundational principles to specialized knowledge tailored to your level of expertise Become part of a global network of learners and professionals who trust Cybellium to guide their educational journey www.cybellium.com **Organic Chemistry, 13e Student Study Guide and Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text **Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show

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