

Unit 5: Physical and Chemical Changes in Matter Study Guide

What is matter?	Anything that has mass and takes up space
Mass is.....	The amount of matter in an object. Measured in grams, kilograms
Volume is.....	The amount of space an object takes up. Measured in milliliters and liters
How would you find the volume of an irregular object, such as a rock, using only a graduated cylinder and water? Describe the process.	Read and record the volume of the water. Drop the rock into the water. Read and record the volume of the rock and the water, then subtract the volume of the water from it.
Describe how particles of matter are arranged in a SOLID. Draw a picture showing this arrangement.	Particles are tightly packed together. They can vibrate, but cannot move out of position.
Describe how particles of matter are arranged in a LIQUID. Draw a picture showing this arrangement.	Particles are further apart, and can flow past one another.
Describe how particles of matter are arranged in a GAS. Draw a picture showing this arrangement.	Particles are much farther apart, rarely coming in contact with one another.
What is density? Use this definition to explain why some object can float and some sink. (textbook p 163)	The amount of mass per unit of volume. If an object is denser than the fluid it is in, it will sink. If it is less dense, it will float.

Study Guide Chemical Changes In Matter Answers

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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AP Biology Study Guide Sundar Nathan,2009-11 Sundar Nathan received a Bachelor s degree in Electrical Engineering from Anna University Chennai India and a Masters degree in Biomedical Engineering from the University of Texas at Austin Working for over a year with a team of talented Phds MPhils and MScs from all over the world Sundar compiled this comprehensive study guide to help students prepare diligently understand the concepts and Crush the AP Bio Test *Study Guide with Selected Solutions* Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone,Theodore O. Dolter,2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

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