

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

### Solutions

#### Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air! (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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Chemistry: Matter and Change • Chapter 15

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## CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

#### Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True \_\_\_\_\_ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False \_\_\_\_\_ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False \_\_\_\_\_ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True \_\_\_\_\_ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True \_\_\_\_\_ 18. Additional solute can be dissolved in an unsaturated solution.

True \_\_\_\_\_ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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# Study Guide Content Mastery Teacher Edition

**Rebecca Barr, Michael L. Kamil, Peter B.  
Mosenthal, P David Pearson**



## **Study Guide Content Mastery Teacher Edition:**

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Minnesota Reading Quarterly ,1972 *Handbook of Reading Research, Volume II* Rebecca Barr,Michael L. Kamil,Peter B. Mosenthal,P David Pearson,2016-11-18 A comprehensive overview of important contemporary issues in the field of reading research from the mid 1980s to mid 1990s this well received volume offers readers an examination of literacy through a variety of lenses some permitting microscopic views and others panoramic views A veritable who s who of specialists in the field chapter authors cover current methodology as well as cumulative research based knowledge Because it deals with society and literacy the first section provides the broadest possible view of literacy The second section defines the range of activities culturally determined to be a part of the enterprise known as literacy The third focuses on the processes that individuals engage in when they perform the act of reading The fourth section visits the environment in which the knowledge that comprises literacy is passed on from one generation to the next The last section an epilogue to the whole enterprise of reading research provides apt philosophical reflection

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