

Chemical energy Study Guide

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I have gone over this study guide with my student for about 30 minutes.

Chemical Vs. Physical

1. What is the definition of a chemical property?

How a substance reacts with other substances. Can only be observed when a new substance is created.

2. Name 3 chemical properties.

Flammability, ability to rust, ability to fizz with acid, changes a strange color

3. What is the definition of a physical property?

Qualities of a substance that can be directly observed, creating a new substance is not necessary.

4. Name 3 physical properties.

Phase, ability to dissolve, mass, density, color, smell, texture

5. Is the ability to dissolve a chemical or physical property?

Physical—when something dissolves it does not change the substance, just breaks it into small pieces

6. What is the difference between a chemical and a physical change?

A chemical change creates a new substance, it is a chemical reaction. Physical changes don't create a new substance

7. Give 3 examples of physical changes.

Breaking something, melting or freezing, dissolving sugar

8. Give 3 examples of chemical changes.

Burning, fizzing, rusting, any chemical reaction

Chemical Reactions

1. Give 4 signs that can tell whether a chemical reaction has occurred (from your notes).

Strange color change, strange change of phase, change in energy, change in smell/taste

2. What is the definition of a reactant? Where do you find reactants in an equation?

A chemical that makes a reaction happen, reactants are on the left side

3. What is the definition of a product? Where do you find products in an equation?

A chemical that is made from the reaction, products are on the right side

Study Guide Energy And Chemical Reactions

Didier Musso



Study Guide Energy And Chemical Reactions:

Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise

Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text *Pass the TEAS V! Complete Study Guide with Practice Questions*

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