

Acids, Bases and Salts

Sour or bitter?

The sour and bitter tastes of food are due to acids and bases, respectively, in them.

Litmus

- Acid turns BLUE litmus RED!
- Base turns RED litmus BLUE!

TRICK - AB Road → Acid turns BLUE litmus RED

Indicators

- An indicator tells us whether the given substance is an acid or a base by the change in its colour!
- For now, we need to know about 3 types of indicators:

A) Natural Indicators

- ◇ Indicators which are obtained naturally are called natural indicators, for example: Litmus, red, cabbage, turmeric, onion, vanilla, clove, etc.

B) Olfactory Indicators

- ◇ Substances which change their odour with acid and base are called olfactory indicators.
- Vanilla, cloves, onion, etc.

Study Guide Acids Bases And Salts Answers

David R. Klein, Laurie S. Starkey



Study Guide Acids Bases And Salts Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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