

NUCLEAR CHEMISTRY WORKSHEET

Name _____ Date _____ Block _____

Most nuclear reactions involve changes in the number of protons and/or neutrons in the nucleus of an atom. These changes are called *transmutations* because an atom of one element is changed into an atom of a different element. During these changes, nuclear particles may be absorbed and/or emitted along with a release of energy.

There are four types of nuclear reactions:

- 1) *natural radioactive decay*: transmutations occur by natural radioactivity (the ability of a nucleus to emit a nuclear particle and energy without external stimulation);
- 2) *artificial transmutation*: transmutations occur by nuclear disintegration caused by external stimulation as a scientist bombards a nucleus with a particle (the addition of another nuclear particle makes the nucleus unstable);
- 3) *fission*: certain nuclei having a large mass are bombarded with special particles that cause the nuclei to split into two nuclei each having a smaller mass;
- 4) *fusion*: nuclei of light elements are combined to form heavier nuclei.

Nuclear reactions are written using symbols in the A_ZX notation. The symbols for the major nuclear particles involved in nuclear reactions are given below.

| | |
|----------------|---|
| Alpha particle | ${}^4_2\text{He}$ (an alpha particle is a helium nucleus) |
| Proton | ${}^1_1\text{H}$ (the most common hydrogen nucleus is a proton) |
| Neutron | ${}^1_0\text{n}$ |
| Electron | ${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ (also called a beta particle) |
| Positron | ${}^0_1\text{e}$ or ${}^0_1\beta$ |
| Gamma ray | ${}^0_0\gamma$ |

The table below contains information about nuclear reactions involving the emission of radiation.

| Type of Reaction | NUCLEAR CHANGES | | |
|---------------------------------|---|-----------------------------|---------------------------|
| | Radiation (Particle) | Effect on the Atomic Number | Effect on the Atomic Mass |
| Alpha emission (α) | ${}^4_2\text{He}$ | decrease by 2 | decrease by 4 |
| Beta emission (β) | ${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ | increase by 1 | no change |
| Positron emission (β^+) | ${}^0_1\text{e}$ or ${}^0_1\beta$ | decrease by 1 | no change |
| Gamma emission | ${}^0_0\gamma$ | no change | no change |
| Electron capture (EC) | ${}^0_{-1}\text{e}$ or ${}^0_{-1}\beta$ | decrease by 1 | no change |

Nuclear Chemistry Review Answers Modern Chemistry

JA Banks



Nuclear Chemistry Review Answers Modern Chemistry:

Decoding **Nuclear Chemistry Review Answers Modern Chemistry**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Nuclear Chemistry Review Answers Modern Chemistry**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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