

Student Exploration: Nuclear Decay

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Objectives: Understand the process of nuclear decay, including alpha, beta, and gamma decay, and how they affect the stability of a nucleus.

Prerequisites: Understanding of basic atomic structure, including protons, neutrons, and electrons, and the concept of a nucleus.

Particle	Symbol	Charge	Approximate Mass
Alpha	α	+2	4.0015
Beta	β	-1	0.0005
Gamma	γ	0	0.0000

- The **alpha particle** is a helium nucleus consisting of two protons and two neutrons. What is the mass number of the alpha particle?
- The **beta particle** is an electron, represented by the symbol β^- . What is the mass number of the beta particle?
- The **gamma particle** is a high-energy photon, represented by the symbol γ . What is the mass number of the gamma particle?

Alpha Decay

Alpha decay occurs when a nucleus emits an **alpha particle**, which consists of two protons and two neutrons. This process is represented by the equation:

$${}^A_ZX \rightarrow {}^{A-4}_{Z-2}Y + {}^4_2\text{He}$$

where X is the parent nucleus, Y is the daughter nucleus, and He is the alpha particle.



Alpha decay is a type of radioactive decay that occurs in heavy nuclei. It is the most common type of decay for nuclei with atomic numbers greater than 82.

- Write the equation for the alpha decay of ${}^{238}_{92}\text{U}$. What is the daughter nucleus?
- Write the equation for the alpha decay of ${}^{234}_{90}\text{Th}$. What is the daughter nucleus?
- Write the equation for the alpha decay of ${}^{226}_{88}\text{Ra}$. What is the daughter nucleus?

Nuclear Decay Explore Learning Exploration Guide Answers

Lei Shi



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