THE MOLE AND AVOGADRO'S NUMBER

Name _____

One male of a substance contains Avogadro's Number (6.02 x 1025) of molecules

How many molecules are in the quantities below?

- 1. 2.0 moles x = 20 mg (x (6.02 x 13 3 molecules) = 1.2 x 10 4 molecules
- 2. 1.5 moles x = 1.5 mole x (6.02 x 13 molecules) = 9.0 x 10 molecules
- 3. 0.75 mole x=0.75 m/ol x (6.02 x 13 molecular)=4.5 x 10 molecular
- 4. 15 moles (6.02×1023 molecula) = 9.0×104 moleculas
- 5. 0.35 mole $\times (602 \times 10^{23} \text{ molecular}) = 2.1 \times 10^{23} \text{ molecular}$

How many moles are in the number of molecules below?

- 1. 6.02 × 1003 × moles = 6.02 × 133 moleulas (1mol 32) = 1.0 mol.
- 2. 1.204 x 1034 X moles : 1.204 x 134 molecula (1 mol 6.02 x 133 molecula) 2.0 molecular
- 3. 1.5 × 10²⁰

 × moler = 1.5 × 10 molegala (1 mol (0.00024 mole)

 × moler = 1.5 × 10 molegala (1 mol (0.00024 mole)

 = 2.5 × 10⁴ molegala
- 4. 3.4 × 10²⁴

 × moles: 8.4 × 18 moley les (1 mol 602× 10° molecules) = 565 moles
- 5. 7.5 × 1010

 × mole = 7.5 × 15 molearles (1mol (6.02 × 15 molearles) = 1.2 × 10 mole

Nuclear Decay Instructional Fair If8766

Puja Mehta

Nuclear Decay Instructional Fair If8766:

This book delves into Nuclear Decay Instructional Fair If8766. Nuclear Decay Instructional Fair If8766 is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Nuclear Decay Instructional Fair If8766, encompassing both the fundamentals and more intricate discussions.

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