

## Balancing and Predicting Nuclear Equations Workshop

### Information: Nuclear reactions and ionizing radiation

A **nuclear reaction** is a change in the composition of the nucleus of an atom. This is not normally considered a chemical reaction, and does not depend on what molecule the atom might be in.

There are three types of nuclear reactions: fusion, fission, and radioactivity. Fusion (combining of nuclei into larger nuclei, such as in stars and the sun) and fission ("splitting the atom," such as in a nuclear reactor) do not concern us much in chemistry.

Some isotopes are radioactive, meaning that their nuclei break down ("decay") and give off particles, "rays," or both. There is no simple way to predict which isotopes are radioactive.

**Table 1: Some types of ionizing radiation produced in nuclear reactions**

Type of Radiation	Symbol	Mass Number	Charge	Relative penetrating ability	Shielding required	Biological hazard
Alpha particle	$\alpha$ , ${}^4_2\text{He}$	4	2+	very low	clothing	none unless inhaled
Beta particle	$\beta$ , ${}^0_{-1}e$	0	1-	low	heavy cloth, plastic	mainly to eyes, skin
Gamma ray	$\gamma$ , ${}^0_0\gamma$	0	0	very high	lead or concrete	whole body
Neutron	${}_0^1n$	1	0	very high	water, lead	whole body
Positron	$\beta^+$ , ${}^0_{+1}e$	0	1+	low	heavy cloth, plastic	mainly to eyes, skin

### Balancing Nuclear Equations Practice

- Complete the following nuclear equations in order to balance them. Identify the type of nuclear change represented by the equation.

Type of Nuclear Change



- Write balanced nuclear equations for each process.

a). The alpha decay of radon-198.

b). The beta decay of uranium -237.

# Nuclear Packet Answers

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