

# How to Pass the Chemistry Regents

| 81 | Fission | Fusion | Structure splits into pieces, resulting mass is converted to energy | 1.13 |
|----|---------|--------|---------------------------------------------------------------------|------|
|----|---------|--------|---------------------------------------------------------------------|------|

29. In which process does a heavy nucleus split into two lighter nuclei?

- (1) fission  
(2) fusion  
(3) electrolysis  
(4) neutralization

30. Given the equation:



Which type of nuclear reaction is represented by the equation?

- (1) fission  
(2) fusion  
(3) beta decay  
(4) alpha decay

30. Given the equation representing a reaction:



Which statement explains the energy source for the reaction?

- (1) Mass is gained due to the conversion of mass to energy.  
(2) Mass is gained due to the conversion of energy to mass.  
(3) Mass is lost due to the conversion of mass to energy.  
(4) Mass is lost due to the conversion of energy to mass.

When uranium-235 nuclei are bombarded with neutrons, many different combinations of smaller nuclei can be produced. The production of neodymium-150 and germanium-81 in one of these reactions is represented by the equation below.



Germanium-81 and uranium-235 have different decay modes. Ge-81 emits beta particles and has a half-life of 7.6 seconds.

82. Explain, in terms of nuclides, why the reaction represented by the nuclear equation is a fission reaction. [1]

30. Given the equations representing reactions:



Which reaction equation is represented by each of these equations?

- (1) Both equations represent fusion.  
(2) Both equations represent fission.  
(3) Equation 1 represents fission and equation 2 represents fusion.  
(4) Equation 1 represents fusion and equation 2 represents fission.

30. Which net change occurs in both nuclear fission and nuclear fusion reactions?

- (1) Mass is converted to energy.  
(2) Energy is converted to mass.  
(3) Small nuclei form a larger nucleus.  
(4) A large nucleus forms smaller nuclei.

# Nuclear Chemistry Regents Questions And Answers

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