

## Unit 3 Study Guide: Atomic Structure and Nuclear Energy

## Section 4.1 Early Ideas About Matter

For each statement below, write true or false.

1. Ancient philosophers regularly performed controlled experiments.
2. Philosophers formulated explanations about the nature of matter based on their own experiences.
3. Both Democritus and Dalton suggested that matter is made up of atoms.
4. Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
5. Dalton's atomic theory stated that matter is mostly empty space.
6. Dalton was correct in thinking that atoms could not be divided into smaller particles.
7. Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
8. Dalton thought that all atoms of a specific element have the same mass.
9. Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
10. Dalton's atomic theory was based on careful measurements and extensive research.

## Section 4.2 Defining the Atom

For each item in Column A, write the letter of the matching item in Column B.

Column A

1. Proposed the nuclear atomic model
2. Determined the mass-to-charge ratio of an electron
3. Calculated the mass of an electron

## Column B

- Thomson
- Millikan
- Rutherford

**Draw and label a diagram of each atomic model.**

5. nuclear atomic model

# **Nuclear Unit Study Guide**

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## **Nuclear Unit Study Guide:**

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## **Nuclear Unit Study Guide** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Nuclear Unit Study Guide**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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