Name Block: Date:

Nuclear Chemistry Worksheet

Identify the following types of nuclear radiation

1. *c 2. * He 3. *γ

Has no mass or charge, almost always follows α or β decay

5. Occurs when a neutron breaks apart and forms a proton

6. Least penetrating nuclear decay

Most damaging nuclear decay to the human body

Nuclear decay that can be stopped by skin or paper.

Nuclear decay that can be stopped by aluminum.

As a result, a proton is transformed into a neutron.

11. The opposite of beta decay

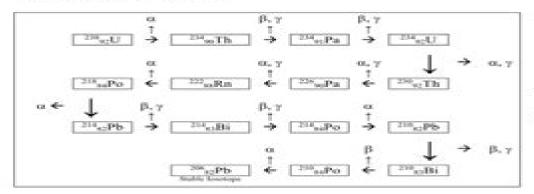
Type of decay isotopes >Z83 will undergo

Complete the following nuclear equations.

18. ____
$$\rightarrow$$
 142 Ba + 91 Kr + 3 1 n

Nuclear Decay Series

The figure maps the radioactive decay of uranium-238 to lead-206. Use the figure to answer the following questions.



19. How many alpha particles are produced as one atom of uranium-238 decays to an atom of lead-206?

20. How many beta particles?

21. Explain why lead-206 is a stable isotope.

Nuclear Chemistry K Answer Key

United States. National Bureau of Standards

Nuclear Chemistry K Answer Key:

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