

NUCLEAR CHEMISTRY WORKSHEET

Part A: Completing the Nuclear Decay Reaction

For each of the atoms listed below, complete the decay reaction by solving for ${}^{\hat{Z}}\hat{X}$ or other missing information. Remember that the mass and protons on each side of the arrow need to equal each other.



Part b: Writing Nuclear Decay Reactions

Write equations for the following nuclear decay reactions. Make sure that both mass numbers and atomic numbers are balanced on each side

11. Decay of polonium-218 by alpha (α) emission

12. Decay of carbon-14 by beta (β) emission)

13. The alpha decay of radon-198

14. The beta (β) decay of uranium-237

15. The alpha decay of plutonium- 243

16. Potassium-40 can decay to form calcium-40. Write the balanced equation for this nuclear reaction.

17. Write a balanced equation for the nuclear reaction in which neon-23 decays to form sodium-23.

18. Write a balanced nuclear equation for the radioactive decay of iron-60 to give cobalt-60.

19. Write a nuclear equation for the following radioactive process; beta emission by argon-37

20. When uranium-234 decays by emitting an alpha particle, it becomes?

Nuclear Chemistry Reading Key Answer

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Nuclear Chemistry Reading Key Answer:

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