- 15. Bromine forms polyatomic ions with structures similar to those of chlorine. Using the chlorine family of polyatomic ions as a model, predict the name of the BrO.1- ion. Cloy is parchlorate; therefore, Brow is per bromate.
- 16. Identify the polyatomic ion in each of these ionic compounds. Write out the name and formula of the ions including their charges.

Ca+2 magnessium Mg+2 ammonium NH4
CO=2 hydroxide OH-1 Chloride CI

# Model 3 – Ternary Ionic Compounds

| Ion Symbols and Charges |   | Chemical Formula  |
|-------------------------|---|---|
| NH,"                    | PO,   | (NH <sub>e</sub> ) <sub>3</sub> PO <sub>4</sub>   |
| Ba <sup>2-</sup>        | NO, <sup>LL</sup>   | Ba(NO <sub>2</sub> ) <sub>3</sub>   |
| NH,1-                   | SO <sub>4</sub> 2-  | (NH <sub>2</sub> ) <sub>3</sub> SO <sub>4</sub>   |
| All*                    | CO,2-   | Al <sub>2</sub> (CO <sub>2</sub> ) <sub>3</sub>   |
| Fe <sup>3</sup> *       | OH:-  | Fe(OH),   |
| K <sup>n</sup>          | NO,1-   | KNO,  |
|                         | NH <sub>4</sub> <sup>1+</sup> Ba <sup>2+</sup> NH <sub>4</sub> <sup>1+</sup> All <sup>2+</sup> Fe <sup>3+</sup> | NH <sub>4</sub> <sup>1-</sup> PO <sub>4</sub> <sup>3-</sup> NO <sub>2</sub> <sup>5-</sup> NO <sub>2</sub> <sup>5-</sup> NH <sub>4</sub> <sup>1-</sup> SO <sub>4</sub> <sup>3-</sup> Al <sup>1-</sup> CO <sub>3</sub> <sup>3-</sup> OH <sup>1-</sup> |

 How are ternary ionic compounds in Model 3 different from binary ionic compounds (NaCl. MgO, CaBr,, etc.) that you've seen previously? Hint: Consider the meaning of the word

Ternary lank compounds contain more than two elements Bisnary ionic compounds contain only two elements.

- \$18. Consider the compound iron(III) hydroxide in Model 3.
  - I How many hydroxide ions (OH") are combined with an iron(III) ion (Fe")?

    There are three hydroxide (OH") for every one install (Fe")
  - to syour answer to part of the only combination of iron(III) and hydroxide that should exist in nature? Explain. Yes. The vatlo of three Fe+ and three OH is the only vatlo that can belone the charge of the two land
  - Consider the compound barium nitrite in Model 3.
    - at. What does the subscripted "2" inside the parentheses of the chemical formula tell you about the compound?

Bar(NO.). The "2" inside of the poventions indicates that there are two oxygen above in the nitrite (NO2") possestance i'en.

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