a) 
$$\times \rightarrow {}^{97}_{41} \text{Nb} + {}^{0}_{-1} \beta$$

b) 
$$^{205}Bi \rightarrow ^{205}Pb + X$$

c) 
$$^{26}_{12}Mg^{1}_{1}p \rightarrow ^{4}_{2}a + X$$

d) 
$$^{238}_{92}U + ^{12}_{6}C \rightarrow X + 6^{1}_{0}n$$

e) 
$$^{59}_{27}$$
Co +  $^{2}_{1}$ H  $\rightarrow$   $^{60}_{27}$ Co + X

f) 
$${}^{97}\text{Tc} \rightarrow {}^{97}\text{Mo} + \times$$

g) 
$$^{53}_{24}Cr + ^{4}_{2}a \rightarrow ^{1}_{0}n + X$$

h) 
$$^{239}Pu \rightarrow ^{235}U + X$$

i) 
$$^{20}80 \rightarrow ^{20}9F + X$$

j) 
$$\times + {}^{4}_{2}He \rightarrow {}^{17}_{8}O + {}^{1}_{1}H$$

k) 
$$^{26}_{12}Mg + X \rightarrow ^{24}_{11}Na + ^{4}_{2}He$$

I) 
$$^{252}_{98}$$
Cf  $\rightarrow ^{142}_{56}$ Ba +  $\times$  + 4  $^{1}_{0}$ n

m) 
$$^{235}_{92}U \rightarrow X + 7^{4}_{2}He + 4^{0}_{-1}\beta$$

# **Nuclear Equation Practice Unit 13 Lesson Answers**

**Christopher Black, Mark Anestis** 

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