

Nuclear Chemistry Practice Sheet

Using your knowledge of nuclear chemistry, write the equations for the following processes:

- 1) The alpha decay of iridium-174

- 2) The beta decay of platinum-199

- 3) Positron emission from sulfur-31

- 4) Krypton-76 undergoes electron capture

- 5) Write the symbols for an alpha particle, beta particle, gamma ray, and positron.

- 6) If the half-life for the radioactive decay of zirconium-84 is 26 minutes and I start with a 175 gram sample, how much will be left over after 104 minutes?

- 7) Why is it difficult to make a fusion reaction occur?

Nuclear Chemistry Practice Problems And Solutions

Rachel S Tattersall



Nuclear Chemistry Practice Problems And Solutions:

CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book

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Choppin,Jan-Olov Liljenzin,Jan Rydberg,Christian Ekberg,2013-09-05 Radiochemistry or nuclear chemistry is the study of radiation from an atomic and molecular perspective including elemental transformation and reaction effects as well as physical health and medical properties This revised edition of one of the earliest and best known books on the subject has been updated to bring into teaching the latest developments in research and the current hot topics in the field To further enhance the functionality of this text the authors have added numerous teaching aids examples in MathCAD with variable quantities and options hotlinks to relevant text sections from the book and online self grading tests New edition of a well known respected text in the specialized field of nuclear radiochemistry Includes an interactive website with testing and evaluation modules based on exercises in the book Suitable for both radiochemistry and nuclear chemistry courses

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