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Name:	Date:	Class period:

Unit: Nuclear Chemistry Test Review



What to study?

- Quizzes: #1
- Homework: worksheets #1-#4
- Notes: Properties of Radiation → Applications of Nuclear Chemistry
- Bell Ringers: Matching, Half-lives, Writing Transmutations

What will the questions be like? What can I expect?

This test will be just like the previous quizzes you have taken.

Matching

Short answer

Multiple choice

Calculations

Sample questions/Important Topics

- Fill in the best answer for each of the following:

 - b) Three types of radiation include ______, _____, and
 - c) During radioactive decay, the ______ will emit particles of radiation in order to become more stable.
 - d) The reactions inside the sun is an example of a ______ reaction.
 - e) If you wanted to block ______ radiation, paper or clothing can be used.
 - Radioactive isotopes, or _______, have different mass numbers but the same atomic number and will decay over time.
 - g) ______ is used to calculate the date of a specimen that is embedded deep within the earth.
 - is the time it takes for half a sample of nuclei to decay into more stable nuclei.
 - i) Fusion reactions release ______ energy than fission reactions.
 - For elements that have an atomic number less than 20, stable nuclei have a neutron:proton ratio equal to _______.
 - k) ______ is the spontaneous emission of radiation by an unstable nucleus.
 - A common isotope used to diagnose thyroid disease is called
 .
 - m) During ______ decay, the atomic number will increase by 1 and the mass number will remain unchanged.
 - n) _____ used alpha particles in his Gold Foil Experiment.

Nuclear Chemistry Test Answers

NARAYAN CHANGDER

Nuclear Chemistry Test Answers:

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