

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Nuclear Decay

**Vocabulary:** alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

The chart below gives the locations, charges, and approximate masses of three **subatomic particles**. The approximate mass of each particle is given in universal mass units (u).

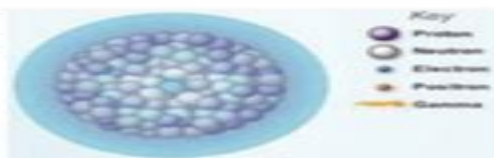
Particle	Location	Charge	Approximate mass
Proton	Nucleus	$1^+$	1 u
Neutron	Nucleus	0	1 u
Electron	Orbitals	$1^-$	0 u

- The **mass number** of an atom is equal to the sum of protons and neutrons in the nucleus. A helium atom has 2 protons and 2 neutrons. What is the mass number of this atom? \_\_\_\_\_
- The **atomic number** of an element is equal to the number of protons in each atom of the element. All helium atoms have 2 protons. What is the atomic number of helium? \_\_\_\_\_

### Gizmo Warm-up

While most atoms are stable, some are **radioactive**, which means that they have a tendency to undergo spontaneous **nuclear decay**. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the Nuclear Decay Gizmo™. On the Gizmo, check that **Alpha decay** and **Uranium** are selected.



- Click **Play** (▶) and then click **Pause** (⏸) when the **alpha particle** is clearly visible. What is an alpha particle made of? \_\_\_\_\_
- Click **Play** and observe. Besides the alpha particle, what else is emitted from the nucleus during alpha decay? \_\_\_\_\_

**Gamma rays** are energetic electromagnetic waves; they are often emitted in nuclear decay.

### Activity A: Alpha decay

Get the Gizmo ready:

- Click **Reset** (↺).
- Check that **Alpha decay** and **Uranium** are selected.



**Question:** How does alpha decay change the nucleus of a radioactive atom?

- Predict:** As you observed in the warm-up activity, an alpha particle consists of two protons and two neutrons. How will the emission of an alpha particle affect the following?
  - The atomic number of the atom: \_\_\_\_\_
  - The mass number of the atom: \_\_\_\_\_
- Calculate:** Turn on **Write equation**. What you see is an equation that shows the original uranium atom on the left. The boxes on the right represent the **daughter product**—the atom produced by radioactive decay—and the emitted alpha particle.
  - In the top left box, write the mass number of the daughter product and press "Enter" on your keyboard. What is this number? \_\_\_\_\_

# Nuclear Decay Gizmo Worksheet Answers

**D. Russell Humphreys, Institute for  
Creation Research**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text box.

## **Nuclear Decay Gizmo Worksheet Answers:**

*Nuclear Decay Scheme Analysis and Characterization Studies of (d, Alpha) Reaction Products* Donald Glenn Gardner, 1957     Determination of Nuclear Decay Schemes by the Coincidence Method James Arthur Cooley, 1952

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