Name

Date _____

Nuclear Chemistry: Identifying Forms of Radiation

Nuclear reactions are quite different than chemical reactions in that they involve a change inside the nucleus rather than an exchange or sharing of electrons. An unstable nucleus decays into a products that are more stable. When an unstable nucleus decays, it releases radiation in the form of alpha particles, beta particles, and/or positrons.

Alpha Decay:	239 Pu	-	235 92 U +	4 2 He
Beta Decay:	60 27 Co —	-	60 28 Ni +	0 -1 β
Positron Emission:	³⁷ Ca	-	37 19 K +	0 +1 β

Directions: Place a check to identify the form of radiation demonstrated by each reaction below.

		Alpha	Beta	Positron
1.	234 Pa → 234 U + 0 91 Pa → 92 U + 0			
2.	222 86 Rn → 218 Po + 4 84 Po + 2 He			
3.	³⁷ κ → ³⁷ Ar + ⁰ β			
4.	99 Tc → 99 Ru + 0 43 Ru + 0			
5.	19 _{Ne} → 19 _F + 0 ₊₁ β			
6.	³² ₁₅ → ³² ₁₆ s + ⁰ ₋₁ β			
7.	238 92 U → 234 90 Th + 4 2 He			
8.	53 Fe 53 Mn + 0 β			
9.	220 Fr			

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