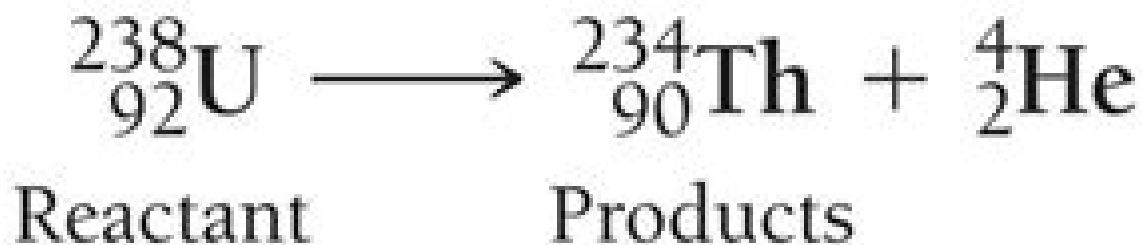


## 2.7 Nuclear Equations and Radioactive Decay

- The general form of a nuclear decay equation is:

*Radioactive nucleus undergoing decay → new nucleus formed + radiation emitted*

- Uranium-238, a radioactive isotope, emits alpha particles when it undergoes radioactive decay. The nuclear equation for this type of decay is:



# Nuclear Equations And Radioactive Decay

**Jerry Marion**



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**General Organic and Biological Chemistry** Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features Chemistry in Quantitative Language Christopher O. Oriakhi, 2021 Chemistry in Quantitative Language is an invaluable guide to solving chemical equations and calculations It provides readers with intuitive and systematic strategies to carry out the many kinds of calculations they will meet in general chemistry This book provides innovative intuitive and systematic strategies to tackle any type of calculations encountered in chemistry Each chapter introduces the basic theories and concepts of a particular topic focusing on relevant equations Worked examples illuminate each type of problem with carefully explained step by step solutions Since chemistry problem can be presented in a number of ways the examples include several versions of each questions To help students understand and retain the procedures the solutions discuss not only what steps to carry out to reach solutions but why The second edition contains additional problems at the end of each chapter with varying degrees of difficulty and many of the original examples have been revised Book jacket EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS CHANG, 2013-01-07 EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS Physics at a Glance Tim Mills, 2008-10-30 This book aims to cover all the GCSE Physics material needed to meet the specifications of the examining boards Edexcel AQA WJEC and OCR both 21st Century Science and Gateway both for single and double awards The content also covers the additional topics necessary for the Physics GCSE single award It is the third book in the series following Healthcare Science T Level: Core Stephen Hoare, Mary Riley, Gemma Roberts, Stephanie France, 2022-10-28 Begin your path to a career in Healthcare Science with this T Level textbook that covers all the core content you will need to understand to be successful in your qualification Develop your understanding of the key principles concepts and theories as well as the skills and confidence to succeed in your written assessment and industry placement Created in partnership with CACHE and written by highly respected authors Stephen Hoare Stephanie France and Gemma Roberts you can feel confident relying on the insights and experience of these experts Get started with short clear summaries and learning outcomes at the beginning of each chapter Track and consolidate your learning using Test Yourself activities throughout each unit Understand and remember key terms using the highlighted definitions Contextualise your learning with case studies reflection tasks and practice points Approach assessment with confidence using the knowledge based questions and scenario based activities for practice alongside model answers for the extended response questions **Chemistry: The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven

Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **Salter's Higher Chemistry** ,1999 This work provides coverage of the content statements in the arrangements for Higher Chemistry organized by the three units in the course Energy Matters the World of Carbon and Chemical Reactions At the start of each unit students are given guidance on what they need to know and understand An Introduction to Chemistry Michael Mosher,Paul Kelter,2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

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