

- Conductivity Test technique: **Without** temperature with **\_\_\_\_\_** (units)
1. **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units)
2. **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units)
3. **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units) **\_\_\_\_\_** (units)

Procedure: Measure "A" for acid, "B" for base, and "C" for salt.

1. **\_\_\_\_\_** **\_\_\_\_\_** **\_\_\_\_\_**
2. **\_\_\_\_\_** **\_\_\_\_\_** **\_\_\_\_\_**
3. **\_\_\_\_\_** **\_\_\_\_\_** **\_\_\_\_\_**

Substance	Acid (A), Base (B), or Salt (C)	Measure with pH (or 1000)	Conductivity (C or 1000)
HClO <sub>4</sub>	A	1	1
HNO <sub>3</sub>	A	10	10
H <sub>2</sub> SO <sub>4</sub>	A	10	10
HClO <sub>3</sub>	A	1	1
HCl	A	1	1
H <sub>2</sub> SO <sub>3</sub>	A	10	10
H <sub>2</sub> SO <sub>4</sub>	A	10	10
H <sub>2</sub> CO <sub>3</sub>	A	10	10
H <sub>2</sub> C <sub>2</sub> O <sub>4</sub>	A	1	1

# Note Taking Guide Episode 1101 Chemistry

**Katrin Zwirglmaier**



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In some sort of inundated with displays and the cacophony of fast connection, the profound power and psychological resonance of verbal beauty frequently fade in to obscurity, eclipsed by the regular onslaught of noise and distractions. However, set within the musical pages of **Note Taking Guide Episode 1101 Chemistry**, a charming perform of literary brilliance that impulses with fresh emotions, lies an remarkable trip waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that exciting opus manuals visitors on a mental odyssey, lightly exposing the latent possible and profound affect embedded within the complex web of language. Within the heart-wrenching expanse of the evocative evaluation, we can embark upon an introspective exploration of the book is central styles, dissect its fascinating publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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