

Recap/Summary

- 1. Study of this _____ multi _____ relationships in a _____ structure _____
- 2. Research on _____ behavioral _____ approaches _____

2. $40g = 40r + 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ in a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

Ex. Problem:

When aluminum chloride reacts with elemental sulfur, aluminum chloride is produced



moles of Al:

$$\frac{2}{3} \times \frac{Al}{S} = \frac{2}{3} \times \frac{Al}{Al_2S_3}$$

2. If you start with 4 moles of Al, how many moles of Al_2S_3 will be produced?

2 moles Al_2S_3 (moles Al_2S_3 = 2 moles Al = 2 moles Al_2S_3)

Note Taking Guide Episode 801 Answers

Georg Christ, Philipp R. Rössner



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