

Background

- 1. Study of the _____ multi _____ relationships in a _____ structure _____
- 2. Research on _____ balanced _____ equations

2. $4\text{Hg} + \text{Cl}_2 \rightarrow 2\text{Hg}_2\text{Cl}_2$

2. Magnesium reacting with a single oxygen atom produces 2 formula units of magnesium oxide

The _____ in a _____ reaction _____ give the _____ ratio _____ for the _____ involved in the reaction _____

E.g. Problem

When aluminum aluminum reacts with elemental oxygen, aluminum oxide is produced



molar ratios

4	Al	3	O ₂
2	Al ₂	3	Al ₂ O ₃
2	O ₂	3	Al ₂ O ₃

2. If you start with 4 moles of Al, how many moles of Al₂O₃ will be produced?

2 moles Al (moles Al₂O₃) and 3 moles O₂ = 2 moles Al₂O₃

Note Taking Guide Episode 80stoichiometry Answers

Michal Rosen-Zvi



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