

Boyle's Law

1. If the _____ of a fixed _____ of gas varies _____ with the _____ of volume _____.
2. _____
3. _____

Charles' Theory and Boyle's Law

1. _____ of a fixed is changed by the _____ of the gas _____ the _____ of the _____.
2. If the gas is _____, the _____ of the volume of fixed _____, assuming _____ the pressure is _____.
3. _____, assuming _____, the _____ of the volume _____ of the _____.
4. _____ of the gas will _____.

Ex. 1. A certain quantity of hydrogen has a volume of 800 ml, at standard atmospheric pressure. After the volume is reduced, it has a pressure of 1.25 atm. What is the pressure initially at 1.00 atm? What is the volume of the hydrogen at this condition?

Ex. 2. A certain quantity of _____, at _____, has a volume of _____ ml. If the pressure is increased until the volume is _____ ml, what is the new pressure, assuming the temperature remains constant?

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SB Merriam



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Current Law Index ,2006

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