

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) Titration with NaOH

a. CH_3COOH volume 25.00 mL conc. 0.100 M

b. NaOH volume 25.00 mL conc. 0.100 M

c. NaOH conc. 0.100 M compound that releases acetate
 CH_3COONa conc. 0.100 M

Record Values of pH and pH for each and pH for each

$\text{pH} = 4.75$ $\text{pH} = 4.75$ $\text{pH} = 4.75$
 $\text{pH} = 4.75$ $\text{pH} = 4.75$ $\text{pH} = 4.75$

Species	Add (M, Base (M, or Salt (M)	Acids with (M) (M or (M)	Conductivity (C or (M)
CH_3COOH	0.100	0.100	0.100
CH_3COO^-	0.100	0.100	0.100
NaOH	0.100	0.100	0.100
NaCH_3COO	0.100	0.100	0.100

Note Taking Guide 1101 Chemistry Answers

Kathleen Armour



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