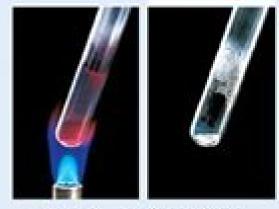
## 2 Decomposition Reaction

Reactants: Generally a single binary compound or a compound with a polyatomic ion Probable Products: Two elements (for a binary compound), or two or more elements and/or compounds (for a compound with a polyatomic ion)

Example: Heating mercury(II) oxide

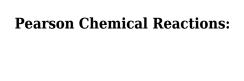


$$2HgO(s) \longrightarrow 2Hg(I) + O_2(g)$$



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