

13 These calculations have used  $L = 6.02 \times 10^{23}$

(a)  $7.22 \times 10^{22}$  (b)  $3.01 \times 10^{23}$

(c)  $1.20 \times 10^{23}$

14 0.53 mol H

15 0.250 mol

16 (a) 262.87 g mol<sup>-1</sup> (b) 176.14 g mol<sup>-1</sup>

(c) 164.10 g mol<sup>-1</sup> (d) 248.22 g mol<sup>-1</sup>

17 189.1 g

18 1.5 mol

19 0.0074 mol Cl<sup>-</sup>

20  $1.83 \times 10^{23}$  C atoms

21 171 g (integer value because no calculator)

22 10.0 g H<sub>2</sub>O

23 2.0 mol N<sub>2</sub> > 3.0 mol NH<sub>3</sub> > 25.0 mol H<sub>2</sub> > 1.0 mol N<sub>2</sub>H<sub>4</sub>

24 (a) CH (b) CH<sub>2</sub>O

(c) C<sub>10</sub>H<sub>12</sub>O<sub>11</sub> (d) C<sub>2</sub>H<sub>6</sub>

(e) C<sub>2</sub>H<sub>6</sub> (f) CH<sub>3</sub>O

25 Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>

26 CoSO<sub>4</sub>·7H<sub>2</sub>O

27 C<sub>11</sub>H<sub>16</sub>N

28 NH<sub>3</sub>

29 6.94 L

30 CdS

31 empirical formula CH; molecular formula C<sub>7</sub>H<sub>8</sub>

32 empirical formula H<sub>2</sub>PO<sub>3</sub>; molecular formula H<sub>4</sub>P<sub>2</sub>O<sub>6</sub>

33 C<sub>10</sub>H<sub>16</sub>N<sub>2</sub>P<sub>2</sub>O<sub>11</sub> for both empirical and molecular formulas

34 C<sub>2</sub>H<sub>3</sub>O

35 Let  $y$  = mass of chalk in grams.

moles of chalk used =  $\frac{\text{mass used}}{M_r(\text{CaCO}_3)} \times$

$y \frac{\text{g}}{100.09 \text{ g mol}^{-1}}$

This is the same as the number of moles of carbon atoms used.

Therefore the number of carbon atoms used =

moles of chalk  $\times$  (6.02  $\times$  10<sup>23</sup> mol<sup>-1</sup>) =

$\frac{6.02 \times 10^{23} y}{100.09}$

36 (a) 2.50 mol

(b) 5.63 mol

(c) 665.5 g

37 (a)  $2\text{C}_2\text{H}_6 + 13\text{O}_2 \rightarrow 8\text{CO}_2 + 10\text{H}_2\text{O}$

(b) 1.59 g

38 4.355 kg

39 (a)  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

(b) 92.8%

(c) CaCO<sub>3</sub> is the only source of CO<sub>2</sub>; all the CaCO<sub>3</sub> undergoes complete decomposition; all CO<sub>2</sub> released is captured; heating does not cause any change in the mass of the other minerals present.

40 (a) 65.2 g

(b) 1.3 g H<sub>2</sub>

41 5.23 g C<sub>2</sub>H<sub>5</sub>Cl<sub>2</sub>

42 253.2 g theoretical CaSO<sub>4</sub>; 77.9%

43 3.16 g ester

44 107 g of C<sub>2</sub>H<sub>6</sub> needed

45 (a) 2.40 mol

(b) 0.0110 mol

(c) 44 mol

46 (a) 35.65 dm<sup>3</sup>

(b) 5.7 dm<sup>3</sup>

47 0.652 dm<sup>3</sup>

48 0.138 mol Br<sub>2</sub> and 0.156 mol Cl<sub>2</sub>, so more molecules of Cl<sub>2</sub>

49 0.113 dm<sup>3</sup>

50 0.28 dm<sup>3</sup>

51 90 kPa

52 16 °C

53 3.0 dm<sup>3</sup>

54 2.8 dm<sup>3</sup>

# Pearson Chemistry Chapters 12 Workbook Answers

**Terry Mahan Buttarro, Patricia Polgar-  
Bailey, Joanne Sandberg-Cook, JoAnn  
Trybulski**

## **Pearson Chemistry Chapters 12 Workbook Answers:**

Introduction to Geochemistry Kula C. Misra, 2012-03-28 INTRODUCTION TO Geochemistry This book is intended to serve as a text for an introductory course in geochemistry for undergraduate graduate students with at least an elementary level background in earth sciences chemistry and mathematics The text containing 83 tables and 181 figures covers a wide variety of topics ranging from atomic structure to chemical and isotopic equilibria to modern biogeochemical cycles which are divided into four interrelated parts Crystal Chemistry Chemical Reactions and biochemical reactions involving bacteria Isotope Geochemistry radiogenic and stable isotopes and The Earth Supersystem which includes discussions pertinent to the evolution of the solid Earth the atmosphere and the hydrosphere In keeping with the modern trend in the field of geochemistry the book emphasizes computational techniques by developing appropriate mathematical relations solving a variety of problems to illustrate application of the mathematical relations and leaving a set of questions at the end of each chapter to be solved by students However so as not to interrupt the flow of the text involved chemical concepts and mathematical derivations are separated in the form of boxes Supplementary materials are packaged into ten appendixes that include a standard state 298 15 K 1 bar thermodynamic data table and a listing of answers to selected chapter end questions

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*E-Book* Terry Mahan Buttaro, Patricia Polgar-Bailey, Joanne Sandberg-Cook, JoAnn Trybulski, 2012-03-01 Written by and for Nurse Practitioners from a unique collaborative perspective Primary Care A Collaborative Practice 4th Edition prepares you to provide care for all of the major disorders of adults seen in the outpatient setting Evidence based content reflects the latest guidelines for primary care of hundreds of conditions including hypertension diabetes and sexually transmitted infections Now in full color the 4th Edition includes chapters on emerging topics such as genetics obesity lifestyle management and emergency preparedness Combining a special blend of academic and clinical expertise the author team provides a practical text reference that promotes a truly collaborative primary care practice Comprehensive evidence based content incorporates the latest standardized guidelines for primary care in today's fast paced collaborative environment Unique A collaborative perspective reflecting the key roles of NPs MDs PAs PharmDs and others promotes seamless continuity of care A consistent format from chapter to chapters facilitates learning and clinical reference value Diagnostics and Differential Diagnosis boxes provide a quick reference for diagnosing disorders and helping to develop effective management plans Physician Consultation icons highlight situations or conditions in which consultation is either recommended or necessary Emergency Referral icons identify signs and symptoms that indicate the need for immediate referral to an emergency facility Co management with Specialists discussions help you provide truly collaborative care in the outpatient setting Complementary and alternative therapies are addressed where supported by solid research evidence

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Rahman,2021-09-22 This book addresses the basic understanding of food contaminants and their sources followed by the techniques to measure food safety and quality It is divided into four parts Part A sources of contaminants in foods their associated health risks and integrated management and alternative options to minimize contaminants Part B Technological assessment of conventional methods and selected advanced methods for the detection identification and enumeration of microbial contaminates Part C Technological assessment of different chemical measurements techniques and Part D Technological assessment of different instrumental techniques to assess sensory properties of foods Food safety is a growing concern due to the increase in food borne illnesses caused by food adulteration excessive use of pesticides use of chemical preservatives and artificial fruit ripening agents microbial contaminations and improper food handling Chemical contaminants in food could be transferred from environmental or agrochemical sources personal care products and other by products of water disinfects In addition microbial food safety can be threatened due to the presence of many pathogens such as Salmonella Escherichia coli Clostridium botulinum Staphylococcus aureus and Listeria monocytogenes in foods Globally strict regulations are imposed to limit the potential contaminants in foods Development of accurate rapid and inexpensive approaches to test food contamination and adulteration would be highly valued to ensure global food safety There are existing processes to ensure safety of food products from chemical and microbial contaminants Apart from the existing measurement technologies varieties of new techniques are also being emerged and these could be potential to ensure food safety and quality In addition to chemical and microbial properties sensory properties such as texture mouth feel flavor and taste are among the most important attributes of food products to ensure their acceptability by consumers Two approaches are available to evaluate sensory properties of food products namely subjective and objective analyses The responses are perceived by all five senses smell taste sight touch and hearing The approach used in sensory evaluation varies depending on the types of foods and the ultimate goal of the testing Sensory attributes are the most important quality parameters after ensuring the safety of foods      *Publishers' circular and booksellers' record* ,1887      **Phi Delta Kappan** ,1916

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