

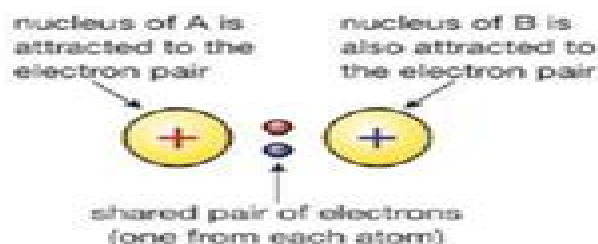
CHAPTER 8: COVALENT BONDING

1.44 Know that a covalent bond is formed between atoms by the sharing of a pair of electrons

- **Covalent Bond:** bond between non-metal and non-metal formed by a shared pair of electrons between the two atoms. It occurs between atoms of non-metal elements.

1.45 Understand covalent bonds in terms of electrostatic attractions

- In a **covalent bond**, a pair of electrons is shared between two atoms. What holds the atoms together is the strong electrostatic attraction between the nuclei (positively charged) of the atoms that make up the bond, and the shared pair of electrons (negatively charged).



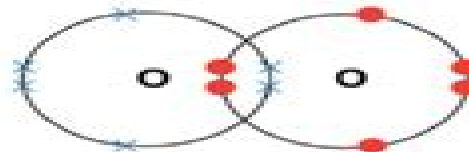
1.46 Understand how to use dot-and-cross diagrams to represent covalent bonds in:

- Diatomic molecules, including hydrogen, oxygen, nitrogen, halogens and hydrogen halides

Hydrogen (H_2)



Oxygen (O_2)



Pearson Chemistry Core Teaching Resources Chapter 8

Covalent Bond

Margaret Irene Lindsay



Pearson Chemistry Core Teaching Resources Chapter 8 Covalent Bond:

Solutions Manual Joseph Topich, 2000-10 Chemistry Rothstein, 1995 **Teaching Chemical Bonding** Margaret Irene Lindsay, 1995 This document presents an instructional strategy for teaching chemical bonding using parables and music Games student interactions and worksheets are included in the lesson plans Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding JRH *Structure, Bonding and Main Group Chemistry* Rod Beavon, Alan Jarvis, 2000 Clear accessible layout and design make texts suitable for a range of student abilities Up to date examples can be used to aid student understanding of important principles and reinforce teaching Texts can be used according to chapters or as self contained dip in resources to suit particular needs Marginal and end of chapter questions to develop student skill and help understanding Typical examination questions enable students to fully prepare for their examinations

Pearson Chemistry Core Teaching Resources Chapter 8 Covalent Bond Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Pearson Chemistry Core Teaching Resources Chapter 8 Covalent Bond**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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