

Answer Key

Chapter 14

Practice Problems

1.
 - a. 61 g CaCl_2 /100 g H_2O
 - b. 90 g CaCl_2 /100 g H_2O
 - c. 5 g $\text{Ce}_2(\text{SO}_4)_3$
 - d. 1 g $\text{Ce}_2(\text{SO}_4)_3$
 - e. 60 g KClO_3
2. 94 atm
3. 2.1 atm
4. 0.852%
5. 7.35 g
6. 492.65 g
7. 26%
8. 0.868 L
9. 0.328*M*
10. 1.09*M*
11. 326 mL
12. 0.802*M*
13. 0.350*m*
14. 0.0300*m*
15.
 - a. water, 0.830; ethanol, 0.170
 - b. water, 0.9853; calcium chloride, 0.0147
 - c. carbon tetrachloride, 0.644; benzene, 0.356
16. BP = 102.06°C;
FP = -7.48°C
17. 2.33*m*
18.
 - a. suspension
 - b. solution or colloid
 - c. colloid
 - d. solution

Chapter 14 Review

19. Solvation is a process in which solvent particles surround solute particles to form a solution. A substance dissolves when attractive forces between solvent and solute particles are strong enough to overcome the attractive forces holding the solute particles together.
20. A saturated solution contains the maximum amount of dissolved solute for a given amount of solvent at a specific temperature and pressure. An unsaturated solution contains less dissolved solute for a given temperature and pressure than does a saturated solution. A supersaturated solution contains more dissolved solute than does a saturated solution at the same temperature.

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James McKeen Cattell, Jaques Cattell



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The Publisher, 1906

The Publishers' Circular and Booksellers' Record, 1925

Chemistry John S. Phillips, Cheryl Wistrom, 2000

The Pearson CSAT Manual 2012 Edgar Thorpe, 2012

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Australian Edition Anita Woolfolk, Kay Margetts, 2012-07-25 Learning and teaching is an integrated process and theory and

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