

Chapter 7 Chemical Reactions

Section 7.1 Describing Reactions

(pages 192–198)

This section discusses the use of chemical equations and how to balance them. It also demonstrates the use of calculations in chemistry.

Reading Strategy (page 192)

Monitoring Your Understanding Preview the Key Concepts, topic headings, vocabulary, and figures in this section. List two things you expect to learn. After reading, state what you learned about each item you listed. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Expect to Learn	What I Learned
Answers may vary. Possible answers: How to balance chemical equations	Answers may vary. Possible answers: An unbalanced equation can be balanced by changing the coefficients.
How to convert from mass to moles	The mass of a substance can be converted to moles by using the molar mass as a conversion factor.

Chemical Equations (pages 192–193)

- Is the following sentence true or false? The new substances formed as a result of a chemical reaction are called products. true
- Circle the letter of each sentence that is a correct interpretation of the chemical equation $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$.
 - Carbon and oxygen react and form carbon monoxide.
 - ☒ Carbon and oxygen react and form carbon dioxide.
 - Carbon dioxide yields carbon and oxygen.
 - ☒ The reaction of carbon and oxygen yields carbon dioxide.
- Is the following sentence true or false? The law of conservation of mass states that mass is neither created nor destroyed in a chemical reaction. true
- Circle the letter of the correct answer. According to the equation $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$, how many carbon atoms react with 14 molecules of oxygen to form 14 molecules of carbon dioxide?
 - 1
 - 7
 - ☒ 14
 - 28
- In the reaction represented by the equation $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$, the mass of carbon dioxide produced equals the total mass of carbon and oxygen that reacted.

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Young Scientist Series ICSE Chemistry 7 Lorraine Mirza, Krishnakali Gupta, *Introduction To Time-dependent Quantum Mechanics With Python* Atanu Bhattacharya, Elliot R Bernstein, 2023-10-18 Computational spectroscopy and computational quantum chemical dynamics is a vast field in physical chemistry Significant part of this field is developed based on the concepts of time dependent quantum mechanics and its numerical implementations This book gives an introduction to the Time Dependent Quantum Chemistry for use with any introductory college university course in optics spectroscopy kinetics dynamics or experimental physical chemistry or chemical physics of the kind usually taken by undergraduate and graduate students in physical chemistry In this book different concepts of time dependent quantum mechanics are systematically presented by first giving emphasis on the contrasting viewpoint of classical and quantum mechanical motion of a particle then by demonstrating the ways to find classical flavour in quantum dynamics thereafter by formally defining the wavepacket which represents a quantum particle and finally by demonstrating numerical methods to explore the wavepacket dynamics in one dimension Along with the analytical theory accompanying Python chapters in this book take readers to a hands on tour with Python programming by first giving them a quick introduction to the Python programming then by introducing the position space grid representation of the wavefunction thereafter by making them familiarized with the Fourier transform to represent the discretized wavefunction in momentum space subsequently by showing the Python based methodologies to express Hamiltonian operator in matrix form and finally by demonstrating the entire Python program which solves the wavepacket dynamics in one dimension under influence of time independent Hamiltonian following split operator approach Rigorous class testing of the presented lecture notes at the Indian Institute of Science GITAM University and at NPTEL platform reveals that physical chemistry students after thoroughly going through all chapters not only develop an in depth understanding of the wavepacket dynamics and its numerical implementations but also start successfully writing their own Python code for solving any one dimensional wavepacket dynamics problem

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