

## Answers

- 55  $M = 133 \text{ g mol}^{-1}$  so gas is Xe
- 56  $90.4 \text{ g mol}^{-1}$
- 57 Helium
- 58  $311 \text{ dm}^3$
- 59 empirical formula and molecular formula =  $\text{SO}_2$
- 60 At higher altitude the external pressure is less. As the air in the tyre expands on heating (due to friction with the road surface), the internal pressure increases.
- 61 (a) Particles are in constant random motion and collide with each other and with the walls of the container in perfectly elastic collisions. The kinetic energy of the particles increases with temperature. There are no inter-particle forces and the volume of the particles is negligible relative to the volume of the gas.
- (b) At low temperature, the particles have lower kinetic energy, which favours the formation of inter-particle forces and reduces gas pressure.  $\frac{PV}{nRT} < 1$
- 62  $\text{NH}_3$  shows greater deviation than  $\text{CH}_4$  due to stronger intermolecular attractions, especially at low temperature.
- 63 B
- 64  $2.81 \text{ g}$
- 65  $4.93 \text{ g}$
- 66  $0.0100 \text{ mol}$
- 67  $0.400 \text{ mol dm}^{-3}$
- 68  $3.1 \text{ cm}^3$
- 69  $0.178 \text{ mol dm}^{-3}$
- 70  $0.0220 \text{ mol dm}^{-3}$ ,  $0.0602\% \text{ HCl}$
- 71  $0.106 \text{ mol dm}^{-3} \text{ Na}_2\text{SO}_4$  and  $0.115 \text{ mol dm}^{-3} \text{ Pb}(\text{NO}_3)_2$ ; assume no side reactions, all  $\text{PbSO}_4$  precipitates
- 72 1217 tonnes

4

- 73 52%  $\text{NH}_3$  by mass; assuming no side reactions occur and gases behave as ideal gases
- 74  $3.20 \times 10^2 \text{ kg}$
- 75  $0.225 \text{ mol dm}^{-3}$  (or round to  $0.23 \text{ mol dm}^{-3}$ )
- 76  $[\text{Na}_2\text{CO}_3] = \frac{VP}{x} \text{ mol dm}^{-3}$   
 $[\text{NaHCO}_3] = \frac{V(P - P)}{x} \text{ mol dm}^{-3}$
- 77  $100 \text{ mol O}_2$ ;  $3.2 \text{ kg}$

## Challenge yourself

- In cold climates, temperature may approach or go below the boiling point of butane so the butane stays liquid even when it is released from the pressure it is under when stored in its canister. This makes it ineffective as a fuel.
- $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ ,  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ,  $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$
- N = 16%, P = 22%, K = 17%
- Many reactions with "useless" by-products could have high stoichiometric yield under optimum conditions, but low atom economy, for example, methanoic acid production:  

$$2\text{NaCOOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{HCOOH} + \text{Na}_2\text{SO}_4$$
 For 100% conversion with stoichiometric reactants, the yield = 100%.  

$$\text{atom economy} = \frac{2 \times 46.03}{(2 \times 68.01) + 98.08} \times 100\% = 39\%$$
- $2\text{NaN}_3(\text{s}) \rightarrow 2\text{Na}(\text{s}) + 3\text{N}_2(\text{g})$   
 $10\text{Na}(\text{s}) + 2\text{KNO}_3(\text{s}) \rightarrow \text{K}_2\text{O}(\text{s}) + 5\text{Na}_2\text{O}(\text{s}) + \text{N}_2(\text{g})$   
 $\text{K}_2\text{O}(\text{s}) + \text{Na}_2\text{O}(\text{s}) + \text{SiO}_2(\text{s}) \rightarrow \text{Na}_2\text{K}_2\text{SiO}_6$  (alkaline silicate glass)
- As  $\text{NaOH}$  dissolves, the separated  $\text{Na}^+$  and  $\text{OH}^-$  ions become hydrated, i.e. they are surrounded by  $\text{H}_2\text{O}$  molecules. This involves breaking the hydrogen bonds between the  $\text{H}_2\text{O}$  molecules in pure water and allows closer packing, which reduces the volume.

# Pearson Chemistry 9 Self Check Activity Answers

**L Darling-Hammond**



## **Pearson Chemistry 9 Self Check Activity Answers:**

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**National Drug Clerk** ,1915 *The Education Index* ,1983 **What Is the Role for Effective Pedagogy In Contemporary Higher Education?** Carl Senior,Dilly Fung,Christopher Howard, Rowena Senior,2018-10-05 The number of students entering into Higher Education HE continues to grow and as such the sector now stands at the threshold of a major shift in its philosophy No longer does the academic prerogative belong to a generation who valued learning for the sake of enlightenment Many contemporary undergraduate students enter their programmes of study with a primary desire to improve their position on the subsequent employability market Universities have been quick to meet this need and institutional offerings have followed suit enabling students to gain experience in a range of additional and subsidiary programmes that focus on the provision of value added benefits Here students are encouraged to develop expertise in a range of topics from entrepreneurship and enterprise to intellectual property and even leadership skills The first round of

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**Paleoclimatology** Gilles Ramstein,Amaëlle Landais,Nathaelle Bouttes,Pierre Sepulchre,Aline Govin,2020-11-09 This two volume book provides a comprehensive detailed understanding of paleoclimatology beginning by describing the proxy data from which quantitative climate parameters are reconstructed and finally by developing a comprehensive Earth system model able to simulate past climates of the Earth It compiles contributions from specialists in each field who each have an in depth knowledge of their particular area of expertise The first volume is devoted to Finding dating and interpreting the evidence It describes the different geo chronological technical methods used in paleoclimatology Different fields of geosciences such as stratigraphy magnetism dendrochronology sedimentology are drawn from and proxy reconstructions from ice sheets terrestrial speleothems lakes and vegetation and oceanic data are used to reconstruct the ancient climates of the Earth The second volume entitled Investigation into ancient climates focuses on building comprehensive models of past climate evolution The chapters are based on understanding the processes driving the evolution of each component of the Earth system atmosphere ocean ice This volume provides both an analytical understanding of each component using a hierarchy of models from conceptual to very sophisticated 3D general circulation models and a synthetic approach incorporating all of these components to explore the evolution of the Earth as a global system As a whole this book provides the reader with a complete view of data reconstruction and modeling of the climate of the Earth from deep time to present day with even an excursion to include impacts on future climate

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