

# ilmkidunya

Student Name	Roll Num	Class Name	Paper Code
Subject Name	Time Allowed	Total Marks	Test Detail
Exam Syllabus	Chapter 2	35	C2

- 1- Circle the correct answer. (8x1=8)**
1. Pressure was kept in discharge tube:  
4 atm (D)  $10^{-4}$  atm (C)  $10^{-2}$  atm (B) 10 atm (A)
2. Number of electrons used in the discharge tube was:  
2 (D) 3 (C) 5 (A) 4 (B)
3. Which one of the following has highest energy value?  
N (D) M (C) L (B) K (A)
4. How many electrons can accommodate in L shell?  
32 (D) 18 (C) 8 (B) 2 (A)
5. N-shell can accommodate electrons:  
50 (D) 32 (C) 18 (B) 8 (A)
6. Gamma rays emit from:  
 $^{238}_{92}\text{U}$  (D) Co-60 (C) Sr-90 (B) P-32 (A)
7. Isotopes of chlorine are:  
1 (D) 4 (C) 3 (B) 2 (A)
8. Which radioisotope is used for the diagnosis of goiter in thyroid gland?  
Iodine-131 (B) Cobalt-60 (A) Phosphorus-30 (D) Strontium-90 (C)
- 2- Answer any 4 short questions. (4x2=8)**
- i. Write the experiments which was done by Sir William Crooks.
- ii. Define Nucleons.
- iii. Explain how anode rays are formed from the gas taken in the discharge tube?
- iv. Why Rutherford considered as the father of nuclear science?
- v. Write the electronic configuration of an element having 11 electrons.
- vi. Magnesium has electronic configuration 2, 8, 2. Why magnesium tend to lose electrons?
- 3- Answer any 5 short questions. (5x2=10)**
- i. How electrons fill the shell?
- ii. Write down the electronic configuration of fluorine and Neon.
- iii. Write down the electronic configuration of  $\text{Al}^{3+}$ . How many electrons are present in its outermost shell?
- iv. What are the isotopes of uranium?
- v. Where nuclear energy used produced from uranium.
- vi. Write two isotopes of Uranium and Chlorine.
- vii. Define Carbon dating.
- viii. How many protons and neutrons are present in chlorine atom  $^{35}_{17}\text{Cl}$ ?
- Attempt any ONE question in detail. (5+4=9)**
- 4(a) Compare the difference between the Rutherford's and Neil Bohr's atomic theories.
- (b) Describe the electronic configuration  $\text{Na}^+$ ,  $\text{Mg}^{2+}$  and  $\text{Al}^{3+}$  ions. Do they have the same number of electrons in the outermost shell?
- 5(a) What is an isotope? Describe the isotopes of hydrogen with diagrams.
- (b) Write notes on isotopes of Carbon and Chlorine.

# Pearson Chemistry Chap 9 Test

**D Siedentop**



## Pearson Chemistry Chap 9 Test:

**The Pearson Complete Guide For Aiee 2/e** Khattar, Lcg Science Chemistry O Lvl , *Pearson Edexcel A Level Chemistry (Year 1 and Year 2)* Andrew Hunt, Graham Curtis, Graham Hill, 2019-07-15 Develop and assess your students knowledge and skills throughout A level with worked examples practical assessment guidance and differentiated end of topic questions in this updated all in one textbook for Years 1 and 2 Combining everything your students need to know for the Pearson Edexcel A level Chemistry specification this revised textbook will Identify the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Student Book Provide support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Improve mathematical skills with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offer plenty of practice with Test yourself questions to help students assess their understanding and measure progress Encourage further reading and study with short passages of extension material Develop understanding with free online access to Test yourself answers and an extended glossary

*Chemistry O Level Mauritius Edition* Pearson Education, Limited, 2007-06-26 Sif: Chemistry S5n Tb J. G. R. Briggs, 2002 **EnvStats** Steven P. Millard, 2013-10-16 This book describes EnvStats a new comprehensive R package for environmental statistics and the successor to the S PLUS module EnvironmentalStats for S PLUS first released in 1997 EnvStats and R provide an open source set of powerful functions for performing graphical and statistical analyses of environmental data bringing major environmental statistical methods found in the literature and regulatory guidance documents into one statistical package along with an extensive hypertext help system that explains what these methods do how to use these methods and where to find them in the environmental statistics literature EnvStats also includes numerous built in data sets from regulatory guidance documents and the environmental statistics literature This book shows how to use EnvStats and R to easily graphically display environmental data plot probability distributions estimate distribution parameters and construct confidence intervals on the original scale for commonly used distributions such as the lognormal and gamma as well as do this nonparametrically estimate and construct confidence intervals for distribution percentiles or do this nonparametrically e g to compare to an environmental protection standard perform and plot the results of goodness of fit tests compute optimal Box Cox data transformations compute prediction limits and simultaneous prediction limits e g to assess compliance at multiple sites for multiple constituents perform nonparametric estimation and test for seasonal trend even in the presence of correlated observations perform power and sample size computations and create companion plots for sampling designs based on confidence intervals hypothesis tests prediction intervals and tolerance intervals deal with non detect censored data perform Monte Carlo simulation and probabilistic risk assessment reproduce specific examples in EPA guidance documents EnvStats combined with other R packages e g for spatial analysis provides the environmental scientist

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**Environmental Statistics with S-PLUS** Steven P. Millard, Nagaraj K. Neerchal, 2000-09-21 A clear comprehensive treatment of the subject Environmental Statistics with S PLUS surveys the vast array of statistical methods used to collect and analyze environmental data The book explains what these methods are how to use them and where to find references to them In addition it provides insight into what to think about before you coll

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**Core Curriculum for Neonatal Intensive Care Nursing E-book** AWHONN, M. Terese Verklan, Marlene Walden, NANN, AACN, 2009-06-24 Perfect as a resource in the field or for exam preparation this authoritative reference from the Association of Women s Health Obstetric and Neonatal Nurses AWHONN includes in depth coverage of the most common neonatal disorders and their management The concise outline format highlights the essentials of each condition including definition etiology pathophysiology signs and symptoms diagnostic tests treatments and outcomes to help you find important information quickly Coverage of key content on the neonatal intensive care nursing certification exam prepares you for test success Information on families ethics legal issues research case management and the transition to home emphasizes the importance of total care of the neonatal patient and family New and updated content on late preterm infants neurologic disorders and ventilatory strategies provide the latest information from the field for more effective patient outcomes

Introductory Chemistry Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking

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**Phlebotomy Essentials, Enhanced Edition** Ruth E. McCall, 2020-06-23

Phlebotomy Essentials Enhanced Seventh Edition provides accurate up to date and practical information and instruction in phlebotomy procedures and techniques along with a comprehensive background in phlebotomy theory and principles

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Biosensors in Precision Medicine Laís Canniatti Brazaca, Juliane Renata

Sempionatto, 2024-02-13 Biosensors in Precision Medicine From Fundamentals to Future Trends covers important topics regarding biomarkers including biomarker discovery validation application in precision medicine the principles of biosensors their use in precision medicine important analytical parameters recent advances in bioreceptors and transduction and more Finally the hottest trends of biosensors in precision medicine including lab on a chip and wearable devices advances towards telemedicine machine learning artificial intelligence and the commerciality of these devices are discussed Provides a timely review of the state of the art developments in biosensors and their applications in precision medicine Presents an evaluation of case studies along with real life examples Addresses recent trends on biosensors for precision medicine

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