

4.1

EXTERNAL ASSESSMENT 2020: CHEMISTRY PAPER 1

Time allowed

- Perusal time — 10 minutes
- Working time — 90 minutes

General instructions

- Answer all questions in this question and response book.
- QCAA-approved calculator permitted.
- QCAA formula and data book provided.
- Planning paper will not be marked.

Section 1 (20 marks)

- 20 multiple choice questions

Section 2 (40 marks)

- 7 short response questions



Section 1 (20 marks)

MULTIPLE CHOICE QUESTIONS

QUESTION 1

A partly filled water bottle is sealed and left on a bench in a room with a constant temperature. After several minutes, it is noted that the water level in the bottle remains constant. In the water bottle, the rate of evaporation is

- (A) less than the rate of condensation.
- (B) greater than the rate of condensation.
- (C) equal to the rate of condensation and equal to zero.
- (D) equal to the rate of condensation but not equal to zero.

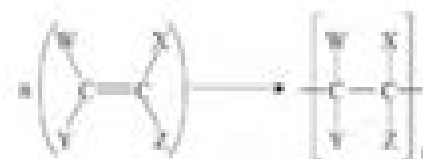
QUESTION 2



Determine which of the following statements is true for the chemical reaction.

- (A) Silver ions are reduced and act as a reducing agent.
- (B) Silver ions are oxidised and act as a reducing agent.
- (C) Magnesium atoms are reduced and act as a reducing agent.
- (D) Magnesium atoms are oxidised and act as a reducing agent.

QUESTION 3



This general chemical equation represents the following type of reaction.

- (A) addition
- (B) hydrolysis
- (C) nitration
- (D) condensation

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Scott C. Dulebohn



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