Answers

Chapter 1

Exercises

- (a) CUCO -+CUC + CO
 - (b) 2Mg + O → 2MgO
 - (c) H SO + 2NaOH → Na SO + 2H O
 - (d) N2 " + __ → 2NH3 "
 - (e) CH + 2O → CO + 2H O
- 2 (a)(2K + 2H O → 2KOH + H

QN OH + 30 → 200 + 3H O

COM 210 → 210CH

46003-x20/203 +

F(eD + 3C -+ 3CO + 2Fe

3 (a)(2C,H_ + 13O, → 8CO, + 10H,O

4866. + 50. → 4NO + 6H,O

3(a) + 8HNO → 3CU(NO) + 2NO + 4H O

6(MD + 2N H -+2N + 10H 0 + 0

460 + N + 150 - 800 + 14H 0 + 2N

- 4 (a) Sand and water: heterogeneous
- Sibilities: tretterogeneous.
- Sociar and water: homogeneous
- S(d) and iron filings: heterogeneous
- E(s)and and water; homogeneous
 - (f) Steel: homogeneous
 - 5 (a)(2KNO_(b)) → 2KNO_(b) + O_(g)
- O(60 O₂(s) + H.SO₂(aq) -+ CaSO₂(s) + CO₂(q) + H Offi

264b) + 2H,O() -→ 2L/OH(ac) + H,(a)

P(dNO,Julisq) + 2NaO(laq) → PbOU(s) +

2NeNO,(eq)

(e) 2C,H,(g) + 9O,(g) → 6CO,(g) + 6H,O()

In the answers, exercises and text there is inconsistency about whether the water formed in combustion reactions is gas or liquid. At room temperature it will ultimately finish up as a liquid. ast shown here.

- X has diffused more quickly, so it must be a lighter gas, its particles have greater velocity than the particles of Y at the same temperature. (Note though that they will both have the same value for average kinetic energy.)
- 7 From the kinetic molecular theory we would expect a solid to be more dense than its liquid. and therefore that ice would sink in water.
- 8 Bubbles will be present through the volume of the liquid. A brown gas is visible above the brown liquid. As the two states are at the same temperature, the particles have the same average kinetic energy and are moving at the same speed. The inter-particle distances in the gas are significantly larger than those in the liquid.
- At certain conditions of low temperature and low humidity, snow changes directly to water vapour by sublimation, without going through the liquid chase.
- 10 Steam will condense on the skin, releasing energy as it forms liquid at the same temperature (E-D) on Figure 1.4). This is additional to the energy released when both the boiling water and the condensed steam cool on the surface of the skin.
- 12 AW 01 EX 12

2

Pearson Chemistry Work Answer Key Chapter 19

David Kirk

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Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation Materials Michael F. Ashby, Hugh Shercliff, David Cebon, 2009-11-20 Materials Engineering Science Processing and Design Second Edition was developed to guide material selection and understanding for a wide spectrum of engineering courses The approach is systematic leading from design requirements to a prescription for optimized material choice This book presents the properties of materials their origins and the way they enter engineering design The book begins by introducing some of the design limiting properties physical properties mechanical properties and functional properties It then turns to the materials themselves covering the families the classes and the members It identifies six broad families of materials for design metals ceramics glasses polymers elastomers and hybrids that combine the properties of two or more of the others. The book presents a design led strategy for selecting materials and processes It explains material properties such as yield and plasticity and presents elastic solutions for common modes of loading The remaining chapters cover topics such as the causes and prevention of material failure cyclic loading fail safe design and the processing of materials Design led approach motivates and engages students in the study of materials science and engineering through real life case studies and illustrative applications Highly visual full color graphics facilitate understanding of materials concepts and properties Chapters on materials selection and design are integrated with chapters on materials fundamentals enabling students to see how specific fundamentals can be important to the design process Links with the Cambridge Engineering Selector CES EduPack the powerful materials selection software See www grantadesign com for information NEW TO THIS EDITION Guided Learning sections on crystallography phase diagrams and phase transformations enhance students learning of these key foundation topics Revised and expanded chapters on durability and processing for materials properties More than 50 new worked examples placed throughout the text The Pearson CSAT Manual 2012 Edgar Thorpe, 2012 The X Site Tim Jones, 2001 Focuses on the Ministry of Supply Factory Rhydymwyn near Mold North Wales aka Valley Works and its important contribution to WWII and the part it might have played had events dictated Physical Chemistry Kurt W. Kolasinski, 2016-09-07 Much of chemistry is motivated by asking How How do I

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