

## 11

## CHEMICAL REACTIONS

## Chapter Test B

## A. Matching

Match each term in Column B with the correct description in Column A. Write the letter of the correct term on the line.

## Column A

- \_\_\_\_\_ 1. a reaction in which an element or compound reacts with oxygen, often producing energy in the form of heat or light
- \_\_\_\_\_ 2. a reaction in which two or more substances react to form a single substance
- \_\_\_\_\_ 3. an equation that indicates only those particles that actually take part in the reaction
- \_\_\_\_\_ 4. a substance that speeds up a reaction without being used up
- \_\_\_\_\_ 5. ions that are not directly involved in a reaction
- \_\_\_\_\_ 6. a reaction in which a single compound is broken down into two or more products
- \_\_\_\_\_ 7. an equation in which each side has the same number of atoms of each element
- \_\_\_\_\_ 8. a reaction in which atoms of an element replace the atoms of a second element in a compound
- \_\_\_\_\_ 9. a list of metals in order of decreasing reactivity
- \_\_\_\_\_ 10. a reaction that involves an exchange of positive ions between two compounds

## Column B

- a. decomposition reaction
- b. activity series
- c. spectator ions
- d. balanced equation
- e. double-replacement reaction
- f. catalyst
- g. combustion reaction
- h. combination reaction
- i. net ionic equation
- j. single-replacement reaction

## B. Multiple Choice

Choose the best answer and write its letter on the line.

- \_\_\_\_\_ 11. In the chemical equation  $2\text{H}_2\text{O}_2(\text{aq}) \xrightarrow{\text{MnO}_2} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ , the  $\text{MnO}_2$  is a:
- a. reactant.  
b. product.  
c. spectator ion.  
d. catalyst.
- \_\_\_\_\_ 12. When the equation  $\text{Mg}(\text{s}) + \text{HCl}(\text{aq}) \rightarrow \text{MgCl}_2(\text{aq}) + \text{H}_2(\text{g})$  is balanced, what is the coefficient for HCl?
- a. 1  
b. 2  
c. 3  
d. 4

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**Michael J. Padilla**



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**Scott Foresman Science** ,2008      *Prentice Hall Chemistry* Antony C. Wilbraham,2008      **Chemistry: The Central Science** Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation      Focus on Physical Science California Edition Michael J. Padilla,2007      *The Pearson Guide to Objective Chemistry for the AIEEE* Singhal Atul,2010-09      *Pharmaceutical Physical Chemistry: Theory and Practices* S. K. Bhasin, Physical chemistry is a compulsory paper offered to all the students of pharmacy There is a dearth of good books that exclusively cover the syllabi of physical chemistry offered to pharmacy courses Pharmaceutical Physical Chemistry Theory and Practices has been designed considering their requirements laid down by AICTE and other premier institutes universities Apart from the theory 20 most common laboratory experiments have been included to make this book a unique offering to the students of pharmacy      *Environmental Microbiology* Eugene L. Madsen,2015-09-28 New and expanded for its second edition Environmental Microbiology From Genomes to Biogeochemistry Second Edition is a timely update to a classic text filled with ideas connections and concepts that advance an in depth understanding of this growing segment of microbiology Core principles are highlighted with an emphasis on the logic of the science and new methods driven discoveries Numerous up to date examples and applications boxes provide tangible reinforcement of material covered Study questions at the end of each chapter require students to utilize analytical and quantitative approaches to define and defend arguments and to apply microbiological paradigms to their personal interests Essay assignments and related readings stimulate student inquiry and serve as focal points for teachers to launch classroom discussions A companion website with downloadable artwork and answers to study questions is also available Environmental Microbiology From Genomes to Biogeochemistry Second Edition offers a coherent and comprehensive treatment of this dynamic emerging field building bridges between basic biology evolution genomics ecology biotechnology climate change and the environmental sciences      *Basic Principles and Calculations in Chemical Engineering* David M. Himmelblau,James B. Riggs,2022-07-27 The 1 Guide to Chemical Engineering Principles Techniques Calculations and Applications Revised Streamlined and Modernized with New Examples Basic Principles and Calculations in Chemical Engineering Ninth Edition has been thoroughly revised

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