CHEMICAL REACTIONS AND EQUATIONS

Objective Questions

Multiple	Choice	Quest	ions
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- 1. Which of the statements about the reaction below are incorrect ? $2PbO(s) + C(s) \longrightarrow 2Pb(S) + CO_2(g)$
 - (i) Lead is getting reduced
 - (ii) Carbon dioxide is getting oxidised
 - (iii) Carbon is getting oxidised
 - (iv) Lead oxide is getting reduced
 - (a) (i) and (ii)
- (c) (i), (ii) and (iii) 2. Fe₂O₃ + 2Al --- Al₂O₃ + 2Fe
- (b) (i) and (iii)
- (d) all of these.

The above reaction is an example of a-

- (a) combination reaction
- (c) decomposition reaction

(b) double displacement reaction

(2024)

- (4) displacement reaction.
- 3. What happens when dilute hydrochloric acid is added to iron fillings? (2022) (a) Hydrogen gas and iron chloride are produced
 - (b) Chlorine gas and iron hydroxide are produced
 - (c) No reaction takes place
 - (d) Iron salt and water are produced.
- 4. In the reaction 2FeCl₂ + Cl₂ --- > 2FeCl₃ Chlorine is a
 - (a) Reducing agent

(b) Oxidizing agent

(c) Catalyst

- (d) Inert medium.
- 5. Rancidity can be prevented in food items by
 - (a) Mixing antioxidant

(b) Keeping in refrigerator

(c) Keeping away from light

- Let All of these.
- 6. Which among the following is not an oxidizing agent?
 - (b) Conc. Sulphuric acid

(a) Oxygen (c) Chlorine

- (4) Hydrogen.
- 7. Which among the following gases are used to keep the oily food items fresh?
 - (a) Carbon dioxide and Oxygen
- (b) Nitrogen and helium

(c) Helium and Oxygen

(d) Nitrogen and Oxygen.

- 8. Chemically rust is-
 - (a) Hydrated ferrous oxide

- (b) Ferric oxide
- (c) Hydrated ferric oxide 9. An exothermic reaction in which glucose combines with oxygen in the cells of
- (d) None of these,
- our body and provides energy-
 - (a) Respiration

(b) Digestion

(c) Excretion

(d) None of these.

<u>Pearson Education Chapter 1chemical Reactions Answer</u> <u>Key</u>

Didier Musso

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Focus on Physical Science California Edition Michael J. Padilla, 2007 Leg Ol Sci Chem Pearson Education, Introductory Organic Chemistry and Hydrocarbons Caio Lima Firme, 2019-08-28 A novel Limited, 2007-10-31 proposal for teaching organic chemistry based on a broader and simplified use of quantum chemistry theories and notions of some statistical thermodynamic concepts aiming to enrich the learning process of the organic molecular properties and organic reactions A detailed physical chemistry approach to teach organic chemistry for undergraduate students is the main aim of this book A secondary objective is to familiarize undergraduate students with computational chemistry since most of illustrations of optimized geometries plus some topological graphs and information is from quantum chemistry outputs which will also enable students to obtain a deeper understanding of organic chemistry **Essentials of Chemical Reaction** Engineering H. Scott Fogler, 2011 Accompanying DVD ROM contains many realistic interactive simulations **Teachers' Use of Visual Representations** Billie Eilam, John K. Gilbert, 2014-07-11 This book examines the diverse use of visual representations by teachers in the science classroom It contains unique pedagogies related to the use of visualization presents original curriculum materials as well as explores future possibilities. The book begins by looking at the significance of visual representations in the teaching of science It then goes on to detail two recent innovations in the field simulations and slowmation a process of explicit visualization It also evaluates the way teachers have used different diagrams to illustrate concepts in biology and chemistry Next the book explores the use of visual representations in culturally diverse classrooms including the implication of culture for teachers use of representations the crucial importance of language in the design and use of visualizations and visualizations in popular books about chemistry It also shows the place of visualizations in the growing use of informal self directed science education Overall the book concludes that if the potential of visualizations in science education is to be realized in the future the subject must be included in both pre service and in service teacher education It explores ways to develop science teachers representational competence and details the impact that this will have on their teaching The worldwide trend towards providing science education for all coupled with the increased availability of color printing access to personal computers and projection facilities has lead to a more extensive and diverse use of visual representations in the classroom This book offers unique insights into the relationship between visual representations and science education making it an ideal resource for educators as well as researchers in science education General Chemistry Wismer, 1999-09-27 visualization and pedagogy *Prentice Hall Chemistry* Antony C. Wilbraham, 2008 Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for

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