

Chapter 1

CHEMICAL REACTIONS AND EQUATIONS

Objective Questions

Multiple Choice Questions

- Which of the statements about the reaction below are incorrect?
$$2\text{PbO(s)} + \text{C(s)} \longrightarrow 2\text{Pb(l)} + \text{CO}_2\text{(g)}$$
 - (i) Lead is getting reduced
 - (ii) Carbon dioxide is getting oxidised
 - (iii) Carbon is getting oxidised
 - (iv) Lead oxide is getting reduced

(a) (i) and (ii) (b) (i) and (iii)
(c) (i), (ii) and (iii) (d) all of these.
- $$\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$$

The above reaction is an example of a— (2024)

 - (a) combination reaction
 - (b) double displacement reaction
 - (c) decomposition reaction
 - (d) displacement reaction.
- What happens when dilute hydrochloric acid is added to iron filings? (2022)

 - (a) Hydrogen gas and iron chloride are produced
 - (b) Chlorine gas and iron hydroxide are produced
 - (c) No reaction takes place
 - (d) Iron salt and water are produced.

- In the reaction $2\text{FeCl}_2 + \text{Cl}_2 \longrightarrow 2\text{FeCl}_3$ Chlorine is a—

 - (a) Reducing agent
 - (b) Oxidizing agent
 - (c) Catalyst
 - (d) Inert medium.

- Rancidity can be prevented in food items by—

 - (a) Mixing antioxidant
 - (b) Keeping in refrigerator
 - (c) Keeping away from light
 - (d) All of these.

- Which among the following is not an oxidizing agent?

 - (a) Oxygen
 - (b) Conc. Sulphuric acid
 - (c) Chlorine
 - (d) Hydrogen.

- Which among the following gases are used to keep the oily food items fresh?

 - (a) Carbon dioxide and Oxygen
 - (b) Nitrogen and helium
 - (c) Helium and Oxygen
 - (d) Nitrogen and Oxygen.

- Chemically rust is—

 - (a) Hydrated ferrous oxide
 - (b) Ferric oxide
 - (c) Hydrated ferric oxide
 - (d) None of these.

- An exothermic reaction in which glucose combines with oxygen in the cells of our body and provides energy—

 - (a) Respiration
 - (b) Digestion
 - (c) Excretion
 - (d) None of these.

Pearson Education Chapter 1 chemical Reactions Key

**Adolfo Iulianelli, Alfredo
Cassano, Carmela Conidi, Konstantinos
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Pearson Education Chapter 1chemical Reactions Key:

Essentials of Chemical Reaction Engineering H. Scott Fogler, 2011 Accompanying DVD ROM contains many realistic interactive simulations

Membrane Engineering in the Circular Economy Adolfo Iulianelli, Alfredo Cassano, Carmela Conidi, Konstantinos Petrotos, 2022-04-12 Membrane Engineering in the Circular Economy Renewable Sources Valorization in Energy and Downstream Processing in Agro food Industry describes the modification of the general concept of waste including waste valorization as added value products that are useful for energy production and biotechnology industries Speaking to the relevance of this new vision the book highlights the fundamentals of membrane operations in the exploitation of renewable sources for energy production and the valorization of agro food waste at the industrial level This book is an excellent resource for researchers biologists membranologists and engineers in chemistry biochemical engineering food sciences and the agro food refinery industry Discusses membrane engineering for agro food wastes transformation into added value products Presents circular and zero waste economy principles pursued by membrane technology and applied to the agro food industry Includes potentialities of agro food wastes for renewable and energy production via membrane operations

Elements of Chemical Reaction Engineering H. Scott Fogler, 2020-08-18 The Definitive Guide to Chemical Reaction Engineering Problem Solving With Updated Content and More Active Learning For decades H Scott Fogler s *Elements of Chemical Reaction Engineering* has been the world s dominant chemical reaction engineering text This Sixth Edition and integrated Web site deliver a more compelling active learning experience than ever before Using sliders and interactive examples in Wolfram Python POLYMATH and MATLAB students can explore reactions and reactors by running realistic simulation experiments Writing for today s students Fogler provides instant access to information avoids extraneous details and presents novel problems linking theory to practice Faculty can flexibly define their courses drawing on updated chapters problems and extensive Professional Reference Shelf web content at diverse levels of difficulty The book thoroughly prepares undergraduates to apply chemical reaction kinetics and physics to the design of chemical reactors And four advanced chapters address graduate level topics including effectiveness factors To support the field s growing emphasis on chemical reactor safety each chapter now ends with a practical safety lesson Updates throughout the book reflect current theory and practice and emphasize safety New discussions of molecular simulations and stochastic modeling Increased emphasis on alternative energy sources such as solar and biofuels Thorough reworking of three chapters on heat effects Full chapters on nonideal reactors diffusion limitations and residence time distribution About the Companion Web Site umich.edu/elements/6e/index.html Complete PowerPoint slides for lecture notes for chemical reaction engineering classes Links to additional software including POLYMATHTM MATLABTM Wolfram MathematicaTM AspenTechTM and COMSOLTM Interactive learning resources linked to each chapter including Learning Objectives Summary Notes Web Modules Interactive Computer Games Solved Problems FAQs additional homework problems and links to Learncheme Living Example

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Table of Contents Pearson Education Chapter 1chemical Reactions Key

1. Understanding the eBook Pearson Education Chapter 1chemical Reactions Key
 - The Rise of Digital Reading Pearson Education Chapter 1chemical Reactions Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Education Chapter 1chemical Reactions Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Education Chapter 1chemical Reactions Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Education Chapter 1chemical Reactions Key
 - Personalized Recommendations
 - Pearson Education Chapter 1chemical Reactions Key User Reviews and Ratings
 - Pearson Education Chapter 1chemical Reactions Key and Bestseller Lists

5. Accessing Pearson Education Chapter 1chemical Reactions Key Free and Paid eBooks
 - Pearson Education Chapter 1chemical Reactions Key Public Domain eBooks
 - Pearson Education Chapter 1chemical Reactions Key eBook Subscription Services
 - Pearson Education Chapter 1chemical Reactions Key Budget-Friendly Options
6. Navigating Pearson Education Chapter 1chemical Reactions Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Pearson Education Chapter 1chemical Reactions Key Compatibility with Devices
 - Pearson Education Chapter 1chemical Reactions Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pearson Education Chapter 1chemical Reactions Key
 - Highlighting and Note-Taking Pearson Education Chapter 1chemical Reactions Key
 - Interactive Elements Pearson Education Chapter 1chemical Reactions Key
8. Staying Engaged with Pearson Education Chapter 1chemical Reactions Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Education Chapter 1chemical Reactions Key
9. Balancing eBooks and Physical Books Pearson Education Chapter 1chemical Reactions Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Education Chapter 1chemical Reactions Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pearson Education Chapter 1chemical Reactions Key
 - Setting Reading Goals Pearson Education Chapter 1chemical Reactions Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Education Chapter 1chemical Reactions Key
 - Fact-Checking eBook Content of Pearson Education Chapter 1chemical Reactions Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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