

8

COVALENT BONDING

Chapter Test B

A. Matching

Match each term in Column B with the correct description in Column A. Write the letter of the correct term on the line.

Column A	Column B
_____ 1. the mixing of several atomic orbitals to form the same number of equivalent hybrid orbitals	a. molecular compound
_____ 2. two or more solid electron-dot formulas that can be written for the same molecule	b. VSEPR theory
_____ 3. a chemical formula that shows the arrangement of atoms in molecules and polyatomic ions	c. van der Waals forces
_____ 4. the weakest attractions that exist between molecules	d. single covalent bond
_____ 5. because electron pairs repel, molecules adjust their shapes so that the valence electron pairs are as far apart as possible	e. resonance structures
_____ 6. a covalent bond in which one pair of electrons is shared between two atoms	f. polar molecule
_____ 7. a molecule in which one end is slightly negative and the other end is slightly positive	g. hybridization
_____ 8. a covalent bond formed between two different atoms in which the bonding electrons are shared unequally	h. structural formula
_____ 9. an electrically neutral group of atoms joined together by covalent bonds	i. polar covalent bond
_____ 10. a covalent bond in which one atom contributes both bonding electrons	j. coordinate covalent bond

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Index Medicus ,1982 Science Citation Index ,1992 Vols for 1964 have guides and journal lists **Biological Science**

Freeman,Scott Freeman,Warren W. Burggren,2002-03 By Warren Burggren University of North Texas Jay Brewster

Pepperdine University Laurel Hester South Carolina Governor s School for Science and Mathematics Rather than repeat

what is covered in the textbook the Student Study Guide will help students study biology and think like a scientist

Introductory chapters on Data Interpretation Looking for Relationships Experimentation and Writing will be illustrated and

developed for the student Each text chapter will then be covered with the goal of reinforcing the ideas mentioned in

introductory chapters and to tie them to appropriate topics within a chapter

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