

## Chemistry 2 Cumulative Exam Review

Which describes how a centrifuge separates a mixture?

It spins a sample to separate based on density.

It spins a sample to separate based on boiling point.

It heats a sample to separate based on density.

It heats a sample to separate based on boiling point. - Answer It spins a sample to separate based on density.

The molar heat of fusion for water is 6.01 kJ/mol. How much energy is released when 36.8 g of water freezes at its freezing point? Use  $q = n\Delta H$ .

6.12 kJ

12.3 kJ

108 kJ

221 kJ - Answer 12.3 kJ

Which characteristic best identifies an Arrhenius base?

It must donate electrons to form a covalent bond.

It must accept protons from an acid.

It must generate hydroxide ions in water.

It must form a conjugate acid after reacting. - Answer It must generate hydroxide ions in water.

Consider the reaction below.



At equilibrium at 600 K, the following are true.

$[\text{CO}_2] = 9.5 \times 10^{-4} \text{ M}$

$[\text{H}_2] = 4.5 \times 10^{-2} \text{ M}$

$[\text{H}_2\text{O}] = 4.6 \times 10^{-3} \text{ M}$

$[\text{CO}] = 4.6 \times 10^{-3} \text{ M}$

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