

11

CHEMICAL REACTIONS

Chapter Test B

A. Matching

Match each term in Column B with the correct description in Column A. Write the letter of the correct term on the line.

Column A

- _____ 1. a reaction in which an element or compound reacts with oxygen, often producing energy in the form of heat or light
- _____ 2. a reaction in which two or more substances react to form a single substance
- _____ 3. an equation that indicates only those particles that actually take part in the reaction
- _____ 4. a substance that speeds up a reaction without being used up
- _____ 5. ions that are not directly involved in a reaction
- _____ 6. a reaction in which a single compound is broken down into two or more products
- _____ 7. an equation in which each side has the same number of atoms of each element
- _____ 8. a reaction in which atoms of an element replace the atoms of a second element in a compound
- _____ 9. a list of metals in order of decreasing reactivity
- _____ 10. a reaction that involves an exchange of positive ions between two compounds

Column B

- a. decomposition reaction
- b. activity series
- c. spectator ions
- d. balanced equation
- e. double-replacement reaction
- f. catalyst
- g. combustion reaction
- h. combination reaction
- i. net ionic equation
- j. single-replacement reaction

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. In the chemical equation $2\text{H}_2\text{O}_2(\text{aq}) \xrightarrow{\text{MnO}_2} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$, the MnO_2 is a:
- a. reactant.
b. product.
c. spectator ion.
d. catalyst.
- _____ 12. When the equation $\text{Mg}(\text{s}) + \text{HCl}(\text{aq}) \rightarrow \text{MgCl}_2(\text{aq}) + \text{H}_2(\text{g})$ is balanced, what is the coefficient for HCl?
- a. 1
b. 2
c. 3
d. 4

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Physical Chemistry Kurt W. Kolasinski, 2016-11-14 Much of chemistry is motivated by asking How How do I make a primary alcohol React a Grignard reagent with formaldehyde Physical chemistry is motivated by asking Why The Grignard reagent and formaldehyde follow a molecular dance known as a reaction mechanism in which stronger bonds are made at the expense of weaker bonds If you are interested in asking why and not just how then you need to understand physical chemistry Physical Chemistry How Chemistry Works takes a fresh approach to teaching in physical chemistry This modern textbook is designed to excite and engage undergraduate chemistry students and prepare them for how they will employ physical chemistry in real life The student friendly approach and practical contemporary examples facilitate an understanding of the physical chemical aspects of any system allowing students of inorganic chemistry organic chemistry analytical chemistry and biochemistry to be fluent in the essentials of physical chemistry in order to understand synthesis intermolecular interactions and materials properties For students who are deeply interested in the subject of physical chemistry the textbook facilitates further study by connecting them to the frontiers of research Provides students with the physical and mathematical machinery to understand the physical chemical aspects of any system Integrates regular examples drawn from the literature from contemporary issues and research to engage students with relevant and illustrative details Important topics are introduced and returned to in later chapters key concepts are reinforced and discussed in more depth as students acquire more tools Chapters begin with a preview of important concepts and conclude with a summary of important equations Each chapter includes worked examples and exercises discussion questions simple equation manipulation questions and problem solving exercises Accompanied by supplementary online material worked examples for students and a solutions manual for instructors Fifteen supporting videos from the author presenting such topics as Entropy Rate Laws Sequestration Electrochemistry etc Written by an experienced instructor researcher and author in physical chemistry with a voice and perspective that is pedagogical and engaging

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