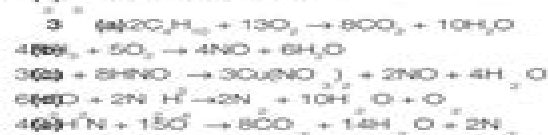
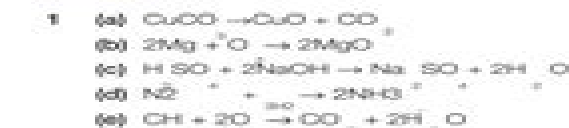


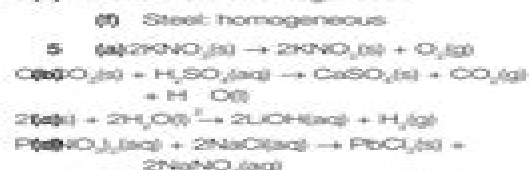
Answers

Chapter 1

Exercises



4 (a) Sand and water: heterogeneous
 (b) Sugar and water: homogeneous
 (c) Oil and iron filings: heterogeneous
 (d) Ethanol and water: homogeneous



In the answers, exercises and text there is inconsistency about whether the water formed in combustion reactions is gas or liquid. At room temperature it will ultimately finish up as a liquid, as shown here.

6 X has diffused more quickly, so it must be a lighter gas. Its particles have greater velocity than the particles of Y at the same temperature. (Note though that they will both have the same value for average kinetic energy.)

7 From the kinetic molecular theory we would expect a solid to be more dense than its liquid, and therefore that ice would sink in water.

8 Bubbles will be present through the volume of the liquid. A brown gas is visible above the brown liquid. As the two states are at the same temperature, the particles have the same average kinetic energy and are moving at the same speed. The inter-particle distances in the gas are significantly larger than those in the liquid.

9 At certain conditions of low temperature and low humidity, snow changes directly to water vapour by sublimation, without going through the liquid phase.

10 Steam will condense on the skin, releasing energy as it forms liquid at the same temperature (E–D on Figure 1.4). This is additional to the energy released when both the boiling water and the condensed steam cool on the surface of the skin.

11 B

12 AW 01_EX_12

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Tewari Akhilesh



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Host-Guest Chemistry Brian D. Wagner, 2020-10-26 This textbook addresses the chemical and physicochemical principles of supramolecular host guest chemistry in solution It covers the thermodynamics and dynamics of inclusion and highlights several types of organic hosts Various applications of host guest chemistry in analytical and environmental chemistry as well as pharmaceutical and chemical industry demonstrate the versatile usability of molecular cages An Introduction To Statistical Thermodynamics Robert P H Gasser, Richards W Graham, 1995-09-26 Statistical thermodynamics plays a vital linking role between quantum theory and chemical thermodynamics yet students often find the subject unpalatable In this updated version of a popular text the authors overcome this by emphasising the concepts involved in particular demystifying the partition function They do not get bogged down in the mathematical niceties that are essential for a profound study of the subject but which can confuse the beginner Strong emphasis is placed on the physical basis of statistical thermodynamics and the relations with experiment After a clear exposition of the distribution laws partition functions heat capacities chemical equilibria and kinetics the subject is further illuminated by a discussion of low temperature phenomena and spectroscopy The coverage is brought right up to date with a chapter on computer simulation and a final section which ranges beyond the narrow limits usually associated with student texts to emphasise the common dependence of macroscopic behaviour on the properties of constituent atoms and molecules Since first published in 1974 as Entropy and Energy Levels the book has been very popular with students This revised and updated version will no doubt serve the same needs **The Mechanisms of Fast Reactions in Solution** Edward Caldin, 2001 Published a few years after the author's death this volume is a sequel to his 1964 book Fast Reactions in Solution the material is entirely new extending investigation beyond now well established fast reaction techniques to consider their contribution to understanding events on the molecular scale After an introductory chapter on origins methods mechanisms and rate constants coverage includes the rates of diffusion controlled reactions mathematical theory of diffusion flash photolysis techniques fluorescence quenching Marcus theory involving proton transfer and group transfer reactions in solutions and electron transfer reactions Annotation copyrighted by Book News Inc Portland OR **Canadian Journal of Chemistry**, 1963 **Construction Science and Materials** Surinder Singh Viridi, 2017-02-06 Construction Science Materials is designed to cover topics studied at levels 2-5 on Construction HND courses and is also suitable for first year undergraduates on construction courses as well as Building surveying Architectural Technology and Quantity Surveying It is an essential text for those who have done no science since their GCSEs Divided into 17 chapters each with written explanations supplemented by solved examples and relevant diagrams to substantiate the text Chapters end with numerical questions covering a range of problems and their answers are given at the end of the book and on the book's website **Longman science Physics 9** Singh, IIT Foundations - Physics Class 9 Pearson, 2009-09 IIT Foundation series is specifically for students preparing for IIT right from school days The series include books from class 8 to

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