$660 \quad 7.22 \times 10^{10}$ 050 3.01 × 10<sup>24</sup> (c)  $1.20 \times 10^{10}$ 14 0.53 mol H 1.5 0.250 mol 16 (a) 262.87 g mol<sup>-1</sup> (b) 176.14 g moh! (c) 164.10 g mot-1 (d) 248.22 g mol-1 189.1.0 117 1.5-mol 1.6 0.0074 mol CF 1.83 × 10<sup>(4)</sup> C atoms 20 171 g (integer value because no calculator) 10.0 g HJO 2121 2.0 mol N, > 3.0 mol NH, > 25.0 mol H, > 1.0 mol N,H, (a) CH 650 CHLO (c) C\_H\_O... (d) C,H, (e) C.H. 00 CHLO Na,5,0,1 26 CoSO\_7HLO 27 C\_H\_N 28 SAR-EL 29 6.94 Lt Costs 3401 34 empirical formula CH; molecular formula C,H, empirical formula H.PO.; molecular formula H.P.O., C.,H.,N,P,O., for both empirical and molecular flormulass. CHLO Let y = mass of chalk in grams. moles of chalk used = mass used

M/(Ca/CO.)

These calculations have used  $L \approx 6.02 \times 10^{10}$ 

This is the same as the number of moles of carbon atoms used.

Therefore the number of carbon atoms used = moles of chalk × (6.02 × 1023 mol-1) =  $9.02 \times 10^{10} V$ 

- \$40 2.50 mol
  - 4ct 665.5 a.

100.09

- 37 (a) 2C,H, + 13O, → 8CO, + 10H,O (b) 1.59 g
- .4.355 kg
- (a) CaCO, -+ CaO + CO,
  - (b) 92.8%
  - (c) CaCO, is the only source of CO,; all the CaCO, undergoes complete decomposition; all CO, released is captured; heating does not cause any change in the mass of the other minerals present.
- 40 (a) 85.2 g
- (b) 1.3 g H.

(b) 5.63 mol

- 5.23 g CJH, CL
- 253.2 a theoretical CaSO.; 77.9%
- 3.16 g ester
- 107 g of C.H., needed
- 45 (a) 2.40 mol (c) 44 mol

Rorm O.E.F.O.O. 6d9

- (a) 35.65 dm\* (b) 5.7 dm<sup>2</sup>
- 0.652 dm<sup>3</sup>
- 0.138 mol Br., and 0.156 mol Cl., so more impleoules of Ct,
- 0.113 dm<sup>3</sup>
- 0.28 dm<sup>3</sup>
- 90 kPs
- 16.10 MILES 201
- 3.0 dm<sup>2</sup>
- 2.8 dm<sup>3</sup>

3

34.500 100,09 g mol\*\*

# **Pearson Chemistry Workbook Answers 8**

**Keith Barnham** 

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partnerships Second are studies empirically examining the implementations of evidence based designs in naturalistic settings and under naturalistic conditions Interestingly the teams conducting these studies are already exemplars of partnerships between researchers and practitioners who are uniquely positioned as in between straddling the two worlds As a result these publications represent both the rigours of research and the pragmatism of reflective practice In forthcoming editions we will add to this collection a third type of publication design profiles These will present practitioner developed pedagogical designs at varying levels of abstraction to be held to scrutiny amongst practitioners instructional designers and researchers alike We hope by bringing these types of studies together in an open access format that we may contribute to the development of new forms of practitioner researcher interactions that promote co design in pedagogical innovation

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